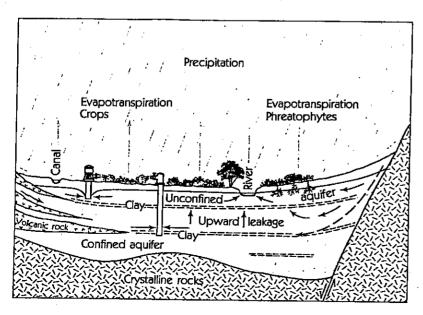
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SALT LAKE COUNTY DEPARTMENT OF PUBLIC WORKS DIVISION OF FLOOD CONTROL & WATER QUALITY

ASSESSMENT OF THE SHALLOW AQUIFER IN SALT LAKE VALLEY



JANUARY 1985

SALT LAKE COUNTY DEPARTMENT OF PUBLIC WORKS
DIVISION OF FLOOD CONTROL AND WATER QUALITY

ASSESSMENT OF SHALLOW AQUIFER QUALITY IN SALT LAKE COUNTY

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I. INTRODUCTION

The Federal Clean Water Act provided planning grants to designated area-wide water quality planning agencies for the purpose of conducting studies which identify the nature and extent of water pollution and its sources in priority drainage basins. The Salt Lake County Council of Governments completed the initial "208" Water Quality Management (WQM) Plan in 1978, and subsequent conditions to its approval began to be addressed by the succeeding grantee, the Salt Lake County Division of Water Quality and Pollution Control. One condition of the WQM Plan to be addressed by the new agency was an assessment of ground-water pollution in the Salt Lake Rasin. Basin.

Early in 1981, Salt Lake County received a grant from the Environmental Protection Agency (EPA) for the purpose of conducting ground-water assessment activities. Preliminary work plans for the study were generated during the winter of 1981, and subsequently revised in the summer of 1981. After an exhaustive period of consultation and training with EPA, the final revised work plan was executed together with a cooperative agreement involving the U.S. Geological Survey (USGS).

Salt Lake County provided staff support for important deep aquifer data gathering activities prior to the cooperative agreement. This support consisted of researching state and local well locations and data for inclusion into the larger USGS deep aquifer network.

The cooperative agreement was executed in an effort to fully integrate and coordinate ground-water study efforts between the Salt Lake County Water Conservancy District, U.S. Geological Survey, Salt Lake County Water Quality, and the Utah Geological and Mineralogical Survey. Initial criticism of duplicated effort was avoided through this agreement, and staff

time and expertise were greatly economized. In view of the work plan contracted between USGS and Water Conservancy, many aspects of the County's EPA-approved work plan were accommodated. For example, the bulk of work needed for further assessment of the deep aquifer was covered in the USGS/Conservancy agreement. Furthermore, the aquifer "interface" outputs described in the County's EPA-approved work plan could only come forth at the conclusion of the Conservancy/USGS water quality tasks. The cooperative agreement between USGS and Salt Lake County Water Quality subsequently covered completion of the shallow aquifer assessment.

The structure of the ground-water work plan and sampling methodology was designed to:

- Provide base datum for ground-water quality which would typify general shallow ground-water quality across Salt Lake Valley.
- Provide data for conditions in proximity to existing non-point pollution sources.

Initial decisions for location of new wells were made on the basis of these goals. It was deemed important to characterize valley-wide shallow aquifer quality and quantity, as well as locate potential leachate plumes requiring further site-specific analysis.

U.S. Geological Survey personnel coordinated closely with Federal and County Water Quality planning personnel in the selection of new well locations.

A. PURPOSE AND SCOPE

The tasks outlined by USGS were consistent with tasks contained within the Salt Lake County workplan. Water level measurement was added as part of the data-gathering effort (see Appendix 1, Table 2). Methodology for site and parameter selection was jointly established through coordination between technical personnel at the Utah State Division of Environmental

Health, U.S. Geological Survey, U.S. Environmental Protection Agency and Salt Lake County Water Quality and Flood Control.

Both field and laboratory Quality Assurance Plans (QAP) were similarly coordinated. Two sample sets were gathered, one by U.S.G.S., one by Salt Lake County. Special care was taken by all parties to insure that specific quality assurance quidelines were met. The decision between EPA and Salt Lake County to obtain an additional sample set resulted in two data reports, which have been combined here as one.

B. SHALLOW AQUIFER MONITORING NETWORK

Existing wells numbered few. Additional wells had to be drilled. A composite factor map was drawn to reflect important variables to be considered for new well locations. These variables included shallow groundwater depths, surface hydrology, land use (with emphasis on hazardous waste sites), soils, recharge/discharge zones and others

C. SHALLOW AQUIFER SAMPLING: QUALITY AND QUANTITY.

From the composite process described above, new wells were located and priority weight established. Certain wells were sampled for organics. Water levels were measured. Water Quality data from site specific investigations were examined.

D. CONCLUSIONS AND RECOMMENDATIONS

Potential Non-point pollution scources - the existence of leachate plumes - are identified. Recommendations for further sampling are made.

It is important that the data base for shallow aquifer conditions be continually expanded. Two data sets do not provide adequate evidence upon which important water use decisions can be based. Long-term data (four - five years) would begin to provide the necessary ingredients to a valid definition of "nature and extent" of ground water pollution. A five year effort

(based on one round of laboratory and field sampling cost for 1983) for the present shallow aquifer network would cost approximately \$15,000. This is an average of only \$3,000 per year. Quarterly samples over a five year period are estimated at about \$60,000 or an average of \$12,000 per year.

with the anticipated reliance on groundwater resources to further supplement future water demands, this cost is relatively small. Potential contamination of deep aquifer reservoirs from shallow aquifer contaminants may present serious diseconomy to our future water resource development policies. Modest expenditures for continued data gathering will vastly improve the quality of critical water policy decision making.

II. EXECUTIVE SUMMARY.

A. INTRODUCTION.

The Salt Lake County Division of Water Quality & Flood Control was designated by the Governor of the State of Utah, under provisions of the Federal Water Pollution Control Act, to be the primary planning & coordinating agency for water pollution control identification and implementation.

This planning responsibility includes the identification of the nature and extent of pollution to classified waters within the Salt Lake Basin, and groundwater requires similar pollution assessments as surface water. Several studies have been performed on the "Deep Confined" aquifer that provides much of the water supply for the County. Little heretofore has been known about the "Shallow Unconfined" aquifer that lies atop the deep aquifer, separated by relatively impermeable clay layers. The new emphasis on hazardous or toxic wastes percolating into the deep aquifer from the shallow aquifer from old landfill or dump areas necessitated more information on the quality and quantity of the shallow groundwater regime. Further, new studies on the availability of the deep aquifer to provide more municipal water supply through "conjunctive use" of ground with surface water implies that pollution may be the single most inhibiting factor to maximizing water use.

The grant-funded shallow aquifer assessment was fully coordinated with the Salt Lake County Water Conservancy District and the United States Geological Survey (USGS). The structure of the County's EPA-approved work plan was designed to provide base data on the general quality of the shallow groundwater resource and begin to collect some information in proximity to suspected non-point pollution sources. Monitoring locations for existing well sampling and new well development were chosen jointly by USGS, EPA, and Salt Lake County personnel.

B. PURPOSE AND SCOPE.

Polluted shallow groundwater may percolate or "discharge" into natural or artificial waterways or may percolate through fractured clay layers into the principal groundwater reservoirs, thus reducing or limiting the present or future use of the water for culinary, recreational, wildlife, fishery, agricultural, or industrial purposes. The orininal scope of the workplan included an examination of the critical "interface" or hydrological relationship between the shallow and deep groundwater regimes. This phase will be delayed for further data acquisition by USGS.

The objectives of the present assesment included measurement of water levels in the shallow aquifer and noting surface water which influences those levels; Identifying soil and land use conditions that influence the permeability of surface water into the groundwater regime; Mapping known water levels in the shallow aquifer and inventorying existing wells and their characteristics; Enlarging the shallow groundwater well network; Monitoring the quality of the shallow aquifer; Report data from site-specific shallow groundwater investigations on-going in the basin; Recommend additional studies of specific locations where shallow aquifer contamination may exist.

The methodology for obtaining shallow groundwater samples dictated a broad spectrum of water quality parameters. The need for an area-wide assessment of quality conditions required that the full range of chemistry be analyzed. All major anions and cations were included, in addition to volatile organics and phenols. At locations where dumps or landfills were known to exist

and leachate contamination suspected, the Environmental Protection Agency placed a priority designation on the well. Analysis was concentrated on the presence of hazardous wastes or toxic contaminants such as trace metals.

In order to insure the integrity and accuracy of sample data, a stringent quality assurance program was required by EPA for both field and laboratory processes and procedures. This involved uniform procedures for sample gathering in the field, preparation of samples for transport, filtering and analysis, along with budgeted provisions for duplicate samples, spiked samples, and blank samples to check laboratory integrity.

C. SHALLOW AOUIFER MONITORING NETWORK.

The majority of wells sampled in the study had to be drilled in cooperation with USGS. Fifty-five (55) wells were ultimately developed for inclusion into the assessment, but only 32 were sampled for water quality. All were measured for water level.

Factors used to determine where new wells should be drilled and which ones should be sampled included: Depth to groundwater conditions (from area soil surveys); Surface hydrology (main creek channels and irrigation canals); General land use (mainly impermeable urban limits and landfills or dumps known to exist throughout the basin); Soil permeability; Recharge, discharge, and perched aquifer zones (identified earlier by USGS). A composite map was produced which displayed these factors together, and well site selection was jointly made by the cooperating agencies.

D. QUALITY OF THE SHALLOW AQUIFER.

Wells were drilled and logged in general terms, and during the one-year duration of the monitoring period, two samples were taken from the entire 32 well network. Because of the time and duration of groundwater movement, sampling should continue seasonally over a period of years-usually three to five-in order to determine the characteristics of the aquifer. Quarterly sampling is planned at some future date. Based on limited knowledge gained from only two samples at each well (a total of 64 sets of data), the following observations can be made:

- 1) A wide range of ph (acid to alkaline content) exists in the shallow groundwater regime, with mean ph substantially more acidic than that found in the deep aquifer. This may indicate general decrease in shallow aquifer quality when compared with the deep aquifer.
- 2) High specific conductivity, total dissolved solids, hardness, and associated cation (metal) concentrations were observed in the Northern and Northwest quadrant of the Salt Lake Valley. Similar conditions were observed at specific wells along the Jordan River. Such conditions may indicate a highly mineralized condition of the shallow aquifer.
- 3) Higher nitrate levels were encountered in the Northern valley quadrants and in areas close to problem land uses, such as landfills, dumps, and animal waste areas.
- 4) Arsenic, Iron, and Manganese were the most frequently measured metals appearing at "higher than background "levels. Arsenic levels occurred most frequently near the International Airport and near the Jordan River between 7800 and 9000 South Streets; Iron occurred in the Northernmost valley region and also near the Jordan river between 7800 and 9000 South Streets; Manganese was found most often in downtown Salt Lake City areas and also near the river between 7800 and 9000 South Streets. The general nature of the shallow aquifer inquiry does not enable the assignment of any specific source to these metal contaminants, nor is there significant statistical support for source identification.

5) Elevated levels of organics--volatiles and phenols--were identified at specific locations during the assessment, but further intensive monitoring and detailed analysis is necessary to determine the risk to human health or impacts to aquatic or terrestrial biota.

During the groundwater study, several site-specific assessments were reviewed and reported. These assessments are being carried out-for the most part-under the requirements of the Resource Conservation & Recovery Act (RCRA) and EPA Superfund programs. They include the AMOCO Oil Company Salt Lake City Refinery Storage Area, the Vitro Uranium Mill Tailings Site, and both old and new Salt Lake City-County Landfill sites. Data for all but Vitro is incomplete, and further analysis precludes sound judgement about known environmental impacts.

E. CONCLUSIONS & RECOMMENDATIONS.

Two samples from a thirty-two well network do not enable the determination of sound scientific conclusions. The thirty-two wells are widely distributed throughout the valley floor and were primarily designed to collect areal data. A few wells located near tailings, dumps or landfills require intensive monitoring, multiple well construction, clustered well design at variable depths, detailed logging of substrate content, frequent sampling, and determination of water level fluctuations over a three to five year period before valid conclusions can be made as to the nature and extent of contamination.

The data gathered during this brief study do show a general decrease in shallow aquifer water quality in the areas of the Salt Lake Valley near the Jordan River and in the Northern to Northwest quadrant. An on-going data gathering effort is in place, funded jointly by Salt Lake County Flood Control & Water Quality and City-County Health Department, that will gather quarterly samples over an undetermined duration. The effort should attempt to establish a base datum of three to five years. Meanwhile, Superfund Regulations administered by EPA will require specific land uses to closely monitor the effects of groundwater leachate on culinary wells, the shallow aquifer, and surface waters.

III.PURPOSE AND SCOPE

The purpose of the shallow aquifer assessment is two-fold. The first is to describe the quality of shallow groundwater on a sub-regional basin scale. The second purpose seeks to define polluted groundwater conditions which are presently or potentially impairing water uses protected under the State of Utah Waste Disposal Code.

Polluted groundwater may discharge to surface streams and impair the uses for which the stream is protected. Local priority surface waters are protected for culinary supply, recreation and aesthetics, aquatic wildlife, agriculture and industry. 7

Groundwater leachate plumes may also be drawn through semi-permeable confining layers into culinary well "cones of depression", resulting in contamination of valuable municipal water supply. 8 Figures 1-7 display basic concepts of the groundwater pollution process.

Present efforts by the U.S. Bureau of Reclamation and State of Utah to classify groundwater reservoirs and their critical boundaries depend on the extent to which pollution plumes are known to exist and on the influence they may have in the future.

The scope of the workplan approved by EPA for Salt Lake County includes an analysis of this critical shallow/deep aquifer interface. However, on-going research by the Salt Lake County Water Conservancy District and USGS will address this interface at a later date. At that time, computer modelling of withdrawal rates and underground transport of leachate into withdrawal zones should prescribe critical limits to this deep/shallow aquifer interface.

A. OBJECTIVES: SHALLOW AQUIFER ASSESSMENT

In coordination with State Health and EPA, and in coordination with

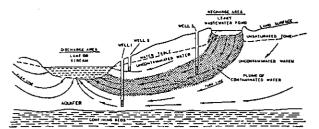


Figure 1: Plumes of contaminated water move along discrete paths between the points of recharge and discharge. The plume is not subject to dilution by the natural ground water. Three monitor wells are shown to indicate that well placement relative to the plume is important in evaluating potential movement of the contaminated water both laterally and vertically.

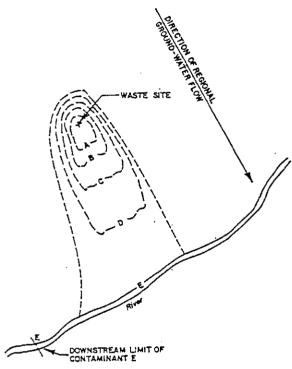


Figure 2: A plan view, representing a specific time in the life of a waste site, shows that specific contaminants (A to E) move different distances in the groundwater system. Contaminant A, which might be biological matter, is attenuated within a short distance; Contaminant E, which might be chloride, is not attenuated.

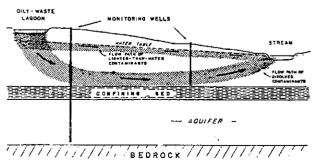


Figure 6: Monitoring wells screened in a lower aquifer or in the plume of dissolved contaminants may not detect lighter-thanwater components of contamination.

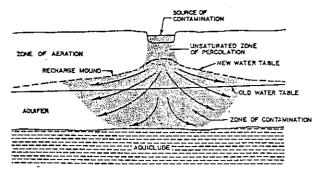


Figure 3: Downward percolation of large volumes of contaminated fluids can result in the development of a recharge mound in the water table beneath a pond or landfill. The mounding of the water table will reduce the thickness of the unsaturated zone beneath the site and, thereby, decrease the attenuation potential of the soil in the vertical dimension.

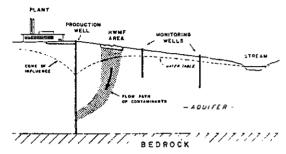


Figure 4: Monitoring wells at lower topographic elevations in the direction of natural ground water flow may not be effective if contaminants are captured by the cone of influence of a nearby pumping well.

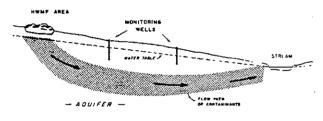


Figure 5: Monitoring wells screened in the upper part of the zone of saturation may not detect contaminants moving through a lower section of the aquifer.

BEDROCK /////////

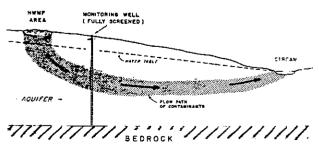


Figure 7: Samples from monitoring wells screened through the entire zone of saturation may yield a mixture of contaminated water diluted with clean water from unaffected sections of the aquifer.

USGS, objectives should address the following data requirements under Phase I inquiry:

- 1. Literature and well data search.
- 2. Characterize physical shallow groundwater system. Emphasize with other environmental variables:
 - a. Map nature and extent of known shallow groundwater conditions.
 - b. Identify surface hydrology and seasonal relationship with groundwater levels, i.e., influence of canal and creek peak discharge.
 - c. Land use effects on possible groundwater and surface water quality and quantity, with emphasis on hazardous waste non-point sources.
 - d. Identify soil conditions with rapid rates of permeability which may transmit pollutants.
- 3. Characterize chemical, biological or organic conditions of shallow groundwater quality and quantity:
 - a. Identify existing well location, ownership, depth, screening, etc
 - b. Tabulate existing data.
 - c. Using factors identified in 2 above, prioritize well monitoring sites for specific parameter selection.
 - d. Monitor water levels (quantity).
- e. Monitor water quality. (Using parameters suggested by EPA)
 Outputs under the Phase I inquiry include:
 - References for previous shallow groundwater monitoring efforts.
 - Composite shallow groundwater factor map displaying priority well locations.

The revised groundwater work plan originally called for integration of shallow and deep aquifer interfaces. This task has been postponed until the USGS deep aquifer assessment is complete and solute transport/withdrawal models run. Specific interface-type outputs will then be available.

Phase II of the Shallow Aquifer Assessment involved evaluation of data generated during Phase I and determination of further sampling needs. Additional chemical and organic analyses were advised for sites downgradient of potential non-point sources, or for those sites possessing high nitrate concentrations. Present EPA priority on high human health risk pollutant sources guided more attention to old landfills or areas meeting hazardous waste definitions under the Resource Conservation and Recovery Act (RCRA).

An outgrowth of the Phase II assessment was examination of groundwater data produced from site-specific monitoring required under RCRA. Examples of this examination include the following site-specific assessments:

- 1) AMOCO waste storage facilities located in northern Salt Lake.
- 2) Rose Park superfund cleanup site, monitored by AMOCO.
- 3) Vitro Uranium Mill Tailings, currently under study by the Federal Department of Energy (DOE).
- 4) Salt Lake County Landfill facilities located just east of the Kennecott tailings pond.

Additional site-specific assessments were initiated during the shallow aquifer assessment. Among these were:

- Kennecott copper mine overburden assessment in Salt Lake Valley.
- 2) Portland Cement facility monitoring, scattered throughout Salt Lake Valley.
- 3) Sharon Steel Mill Tailings superfund assessment
 Outputs of the Phase II Shallow Aquifer Inquiry include:
- 1- Description of selection criteria employed by Salt Lake County, State of Utah and EPA in determining priority well selection.
- 2- Listing of well characteristics.
- 3- Mapping, which displays the occurrence of particular priority

parameters.

4- Implementation of additional shallow well sampling.

Interpretation of sample data must be limited to two data sets. Further sampling of all parameters is necessary for construction of a reliable data base for use in future withdrawal and solute-transport models. However, some obvious problem areas merit concentrated study at a future date.

B. METHODOLOGY

Consistent with the purpose and objectives of the study, two levels of method for site selection and parameter sampling were pursued. Need for an area assessment of typical shallow groundwater quality required broad parameters, while potential site-specific non-point source areas (mainly potential hazardous waste sites), required more specific analysis. Table 1 shows the wells prioritized by the project team, together with designations for anions (mainly nutrients such as nitrates and phosphates), cations (heavy mercury), volatile organics (such metals such cadmium and as dichloroethane, chloroform) and phenols. Hazardous waste sites are typically associated with phenols and organics where benzenes, halomethanes and trihalomethanes are prevalent. Heavy metals are typically linked with landfills, mill tailings or other mining related activities. Anions are associated with biological waste such as wastewater disposal, fertilizers or animal waste concentrations.

Selection of "priority" wells required concurrence with EPA objectives, and both volatile organics and phenols were sampled at these sites in addition to standard cations and anions. This method was intended to meet non-point source identification requirements of the Clean Water Act. New wells should periodically be added to survey other non-point sources, such as stormwater detention basins and sand and gravel pits.

TABLE ONE

SALT LAKE COUNTY SHALLOW AQUIFER ASSESSMENT

SECOND ROUND SAMPLING, PARAMETER, QUALITY ASSURANCE & COST SCHEDULE: BY

S. F. JENSEN & T. WAY 6/22/83

Well US	Identificat AS	ion # Î	EPA*	Water C ANIONS	uality Pa CATIONS	rameters ORGANICS (VOLATILE)	PHENOLS
		<u></u>	<u></u>			(VOLATILE)	
B-1-1	9 ADC 26 BAD		18	X X	X X	χ	Х
	26 CDA		10	X	X	^	^
	32 CCD		3	Х	χ	X	Χ
5 1 0	35 DCB		1	X	χ	v	Χ
8-1-2 C-1-1	34 AAB 2 DCA		1	X Y	X X	Х	٨
C-1-1	4 DDB		4**	X X X X X	X	Χ	Χ
	11 BAC			Χ	χ		,
	15 CAA		044	X	X	v	v
	24 CDC 26 DCA		8** 9**	X,	X X	X X	X X
	28 CAB		,		X	X	χ̈́
	30 ACA			X X X X	Χ	χ	Χ
	31 ABB	,		X	X	X	X
C - 2 - 1	12 BDA		11 12*	X	X X	X	X X
	14 DBD 15 ABC		13		X	X	χ
	26 ABB	•	14*	X X X	Χ	X	X
	34 DDA-3			Χ	Χ		
	35 BAA 35 BAB		15*	X	X	Χ	Х
C - 3 - · 1	35 BAB 1 BBC		16*	X X	X X	Χ	Χ
C 5-1	3 ACC		10	. X	Χ	^	
	34 AAA			Χ	Χ		
D-1-1	18 DAD	•		X	X X		
D-2-1	31 DBA 8 BBA		•	X Y	X		
D-3-1	5 CDB		17	x	X	Χ	Χ .
	6 BCB			X	Χ		
	31 CDA			X X X X X	X X X		
	32 AAA			Х	Х		
TOTALS	32		13	32	32	16	16
	*EPA 1st	Priority W	ells	•			
	**EPA 2nd	Priority W	ells	ssurance :	Samnlac		
		<u> </u>	alicy A.	SSUI GILCE	Jamp res		
Field	Blank			1	1	1	1
Duplic	ate			3	3	3	3
Blind	QC (spike)			1	1	0	0
GRAND	TOTALS			37	37	20	20

C. QUALITY ASSURANCE

A Quality Assurance (QA) component of the study was required for both laboratory and field procedures. Salt Lake County followed Interim Guidelines and Specifications for Preparing Quality Assurance Project Plans (December, 1980), 10 and reviewed QA specifics with EPA staff to insure their consistency with The Quality Assurance Program Plan for EPA Region VIII (January, 1982). 11 The goal of the EPA Quality Assurance Program is to generate data that is scientifically valid, defensible and of known precision and accuracy. 12 USGS Laboratory QA has been developed in coordination with EPA involvment, and the State of Utah QA program has been approved by EPA as consistent with Region VIII EPA plans. 13 First Round samples were processed in the Denver Regional Laboratory of the U.S. Geological Survey. Second Round samples were processed by the Utah State Laboratory in Salt Lake City.

Field QA procedures were drafted jointly by USGS and Salt Lake County Water Quality personnel. EPA approved the procedures consistent with Region VIII policy prior to initiation of Round II sampling. Outlined Field procedures are attached as Figure 8.

DRAFT FIELD QUALITY
ASSURANCE PROGRAM: SALT LAKE COUNTY
SHALLOW AQUIFER ASSESSMENT FIELD SAMPLING PROCEDURES

June 27, 1983

The following description provides step-by-step procedures performed during field sampling for the Salt Lake County Shallow Aquifer Assessment. Three components are cover: (1) Water level measurement; (2) Well pumping and evacuation; (3) water quality sampling. The sampling description includes specific procedures for obtaining anions and cations, volatile organics and phenol samples. All data is recorded on the Water Quality Field Report Form.

I. Water Level Determination

- A. Equipment: Peristaltic pump, 25' plastic pump tubing, metal tape measure, 3 gallon capacity plastic bottle.
 - 1. Extend metal tape through hole in well cap approx. 6'. Remove tape and measure depth of water surface from top of well cap.

2. Extend metal tape into well to bottom or depth of sand of infill. Remove tape and calculate water depth.

3. Calculate well casing volume by multiplying depth X .16 (gallons/ft. of water). Pump at least two well casing volumes prior to sampling.

II. Well Pumping/Evacuation

- A. Equipment: # gal. plastic container, lead-weighted tubing, peristaltic pump, 200 cc. plastic beakers, conductivity meter, centigrade thermometer.
 - 1. Extend lead-weighted plastic tubing midway between surface water level and bottom of well or to top of sand or infill.
 - 2. Begin pumping with peristaltic pump into 3 gallon plastic container. (Container is used for measuring purposes)
 - 3. While pumping, collect first conductivity and temperature sample into 200 cc. plastic beaker.
 - 4. Record water temperature on water quality field sheet.
 - 5. Record conductance reading on water quality field sheet.
 - 6. Conductivity meter is calibrated prior to each sampling day against prepared standards as described by manufacturers specifications(ref.).
 - 7. Using conversion tables (ref.) correct conductivity reading to 25°C.
 - 8. During pumping of two to three well casing volumes, prepare equipment and materials for water quality sampling.

III. Water Quality Sampling

A. Equipment: Same as I & II, 0.45 micron standard filter, filter aparatus, pH meter, pH buffer, sample bottles labeled and prepared with preservatives, cooler w/5lbs. ice, 1 gal. distilled water.

B. Filter Preparation:

- 1. Rinse filtering apparatus thoroughly with distilled water.
- 2. Install cellulose filter on filter appartus and tighten. Do not touch filter with finers

C. pH Measurement:

- 1. Field calibrate pH meter according to manufacturers specifications to pH 7 and pH 10. (ref) pH probe is stored in distilled water.
- 2. Draw second conductivity sample into 200 cc. beaker. Measure temperature, take conductivity reading, and adjust to 25°C.
- 3. Take preliminary measurement.
- 4. Draw third conductiveity sample, measure temperature and determine conductivity. If temperature and conductivity have been constant (i.e., representative of aquifer conditions rather than well conditions) proceed with Part D. If not, continue pumping and repeat Step C-4 until successive conductivities and temperatures are constant.

D. Sampling-Anions/Cations:

- 1. While pumping into 200 cc. beaker, take final pH temperature and conductivity readings. Record on field sheet. Record Well number/location, date, time, filtered or unfiltered, and acid treatment (if not for trace metals) on label outside bottle.
- 2. Rinse untreated bottle thoroughly with sample water.
- Draw sample into bottle and replace cap (untreated, unfiltered sample).
- 4. For filtered samples, pump one sample volume through filter and then draw sample into acid prepared bottle. Watch tube and filter to avoid fouling filter with sediment or other possible debris. If fouling occurs, replace filter and resample.

E. Sampling-Volatile Organics:

- 1. Attach 100 ml. volumetric pipette to end of tubing and insert to center of water column.
- 2. Draw at least 300 ml. water through pipette then turn pump off.
- 3. Drain pipette into septum sample bottle filling from bottom up to avoid air contamination. Avoid exposure of sample to oxygen or plastics to the max.
- 4. Overfill bottle (inverse miniscus). Refit cap and wrap with foil to prevent organic destruction.
- 5. If any air is present in bottle (invert and watch for bubbles), discard and repeat sampling procedure.
- 6. Take duplicate sample as in E-1 through E-5.

F. <u>Sampling-Phenols</u>:

- 1. Attach string to pipette so that the pipette hangs vertically.
- 2. Insert into well casing and allow to fill with water using string as bailing retreiver.
- 3. Remove and drain pipette into sample bottle containing preservatives Copper Sulfate and Sulfuric Acid.
- 4. Repeat until adequate sample volume collected (1 liter).

G. <u>General</u>:

- Ice all samples after collection.
- 2. Fill out laboratory sheets.
- 3. Deliver samples and lab sheets to Utah State Department of Health Laboratory, 44 Medical Drive, Salt Lake City, Room 127 before 4:30 each sample day.

SALT LAKE COUNTY FLOOD CONTROL & WATER QUALITY

WATER QUALITY FIELD REPORT

Date Stat	Time ion Number	Sample#	Project Name	
Colle	ected By		_	
		FIELD MEASI	IREMENTS	
AIR	TEMPERATURE:	OC		
WATE	R LEVEL/CASING VO Water Level: He Well Depth:	ft cut_	ft. = Depth to Waterft.	
	Casing Volume:	Well Depth ft.	Depth to Water ft. = Depth of Water ft.	
		Depth of Water	ft. x Constant = Volume	_gal
SPEC:	IFIC CONDUCTANCE Measurement *1.	(Meter Type and #oc =	=umho @25°C	_).
	2.	umho @oc =	umho @250C	
	3.	umho @oc :	=umho @25oC	
	4.	umho @oC:	=umho @250C	-
	Specific Conduc (*Use additio	tance: umho @ nal pages if more than	oC =umho @250C n 4 preliminary measurements taken).	
	pH (Meter Type Measuremen	and #). t *1Units @_	oC	
		2Units @	oc	
,		3Units @_	oc	
		4Units @	oC	
	pH: (*Use additio	Units @ nal pages if more than	oc n 4 preliminary measurements taken).	
		SAMPLES COL	ECTED	
TYPE	OF SAMPLE TAKEN Anions Cations		ALITY ASSURANCE SAMPLES: Split Sample Replicate Sample	
	Volatile Organi Phenols Others (Specify		Blank Sample Spike Sample	
0BSE1	RVATIONS AND REM	ARKS:		

IV SHALLOW AQUIFER MONITORING NETWORK

Some shallow aquifer wells existed prior to the cooperative study. The majority of the wells had to be drilled during the spring, summer, and early autumn of 1982. Utilizing assistance from Salt Lake County Flood Control and Water Quality, the U.S. Geological Survey drilled approximately 55 wells for observation of both quantity and quality conditions. The need to establish control over flow quantity and direction in the shallow aquifer required extensive water level data throughout Salt Lake Valley. Of the 55 stations, all were measured for water levels. Only 32 stations were selected for water quality analysis.

Location for the wells had to meet two basic criteria:

- To obtain a characterization of general shallow groundwater quality.
- 2) To target potential non-point source contaminant plumes.

Haphazard location of these wells was to be avoided. Factors which would facilitate work plan objectives were separately described and composited to arrive at optimum well locations. These factors included:

- 1) Depth to shallow groundwater.
- Surface hydrology
- 3) General land use Emphasizing urban, rural, wetland, and waste disposal zones
- 4) Soil permeability
- 5) Recharge, discharge, and perched aquifer zones.

Based on analysis of these conditions the location and anticipated characteristics of the shallow wells could be better understood.

A. DEPTH TO GROUNDWATER

Shallow groundwater conditions to a maximum depth of 60" were documented by the Soil Conservation Service (SCS) in the Salt Lake Soil Survey as early

as 1966. 14 Depth increments of 10" (0 - 30", 30" - 40", 40" - 50", 60" or greater) were estimated at locations occurring mainly within bottomlands or in proximity to surface water features or floodplains.

The Utah Geological and Mineralogical Survey (UGMS) incorporated SCS groundwater typology into a larger depth scale in 1982. ¹⁵ Depth increments at 0' - 5', 5' - 10', 10' - 20', and over 20' were delineated by UGMS, in addition to zones over 100' in depth and the occurrences of bedrock outcrops.

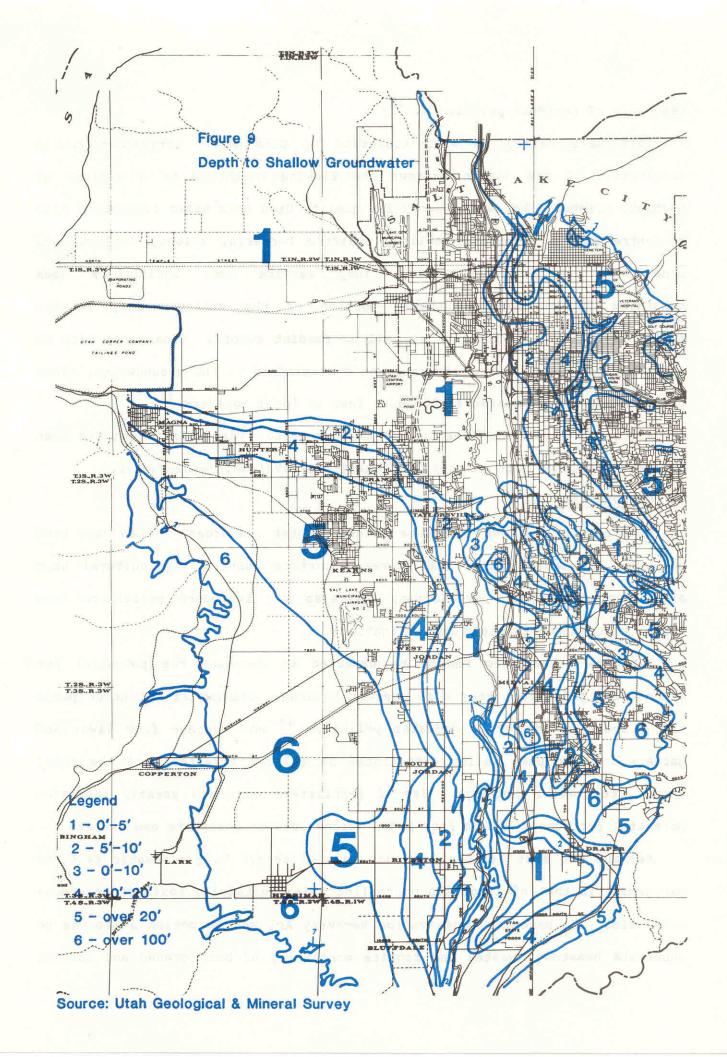
Figure 9 displays the incidence of shallow groundwater locations and relative depths. Variations of groundwater depth in the Class 1 zone (0' - 5') can be determined from soil mapping units in the Salt Lake Soil Survey.

B. SURFACE HYDROLOGY

Salt Lake Valley is intersected by no less than 16 perennial and intermittent freshwater streams. City Creek, Red Butte Creek, Emigration Creek, Parleys Creek, Mill Creek, Big and Little Cottonwood and Dry Creek are the principal perennial drainages to Salt Lake Valley from the Wasatch to the east. Butterfield, Midas, Bingham, and Barneys Creeks are the main intermittent drainages from the Oquirrh Range to the west. The Jordan River and Surplus Canal confluence these drainages at the valley bottom. Coon and Kersey Creeks flow directly into Great Salt Lake.

Eastside drainages are high quality streams protected for recreation, aesthetics, fishing, aquatic wildlife, and irrigation. They discharge over 151,000 acre-feet of mean annual inflow to the Jordan River and recharge 20,000 acre-feet per year to the principal confined aquifer. Pollutants entering these streams from shallow aquifer contamination could impair not only surface water use, but municipal sources as well.

Estimated runoff from unclassified intermittent streams draining the Oquirrhs is about 7000 acre-feet per year. Bingham and Butterfield Creeks are intercepted by mine tailings ponds which imply potential for surface



transport of residual pollutants.

Salt Lake Valley is also traversed by nine major irrigation canals originating at the Jordan Narrows and flowing northward to a variety of terminus points. The canals carry low quality Utah Lake water possessing high concentrations of dissolved solids, coliform bacteria, anions, cations, and occasional pesticides/herbicides. They service some agriculture uses remaining in the south-southwest portion of the valley. They receive irrigation tailwater pollution as well as feedlot runoff. Canals are also an important potential source of non-point contaminants to the groundwater, since they recharge an estimated 48,000 acre feet to local aquifers. 17

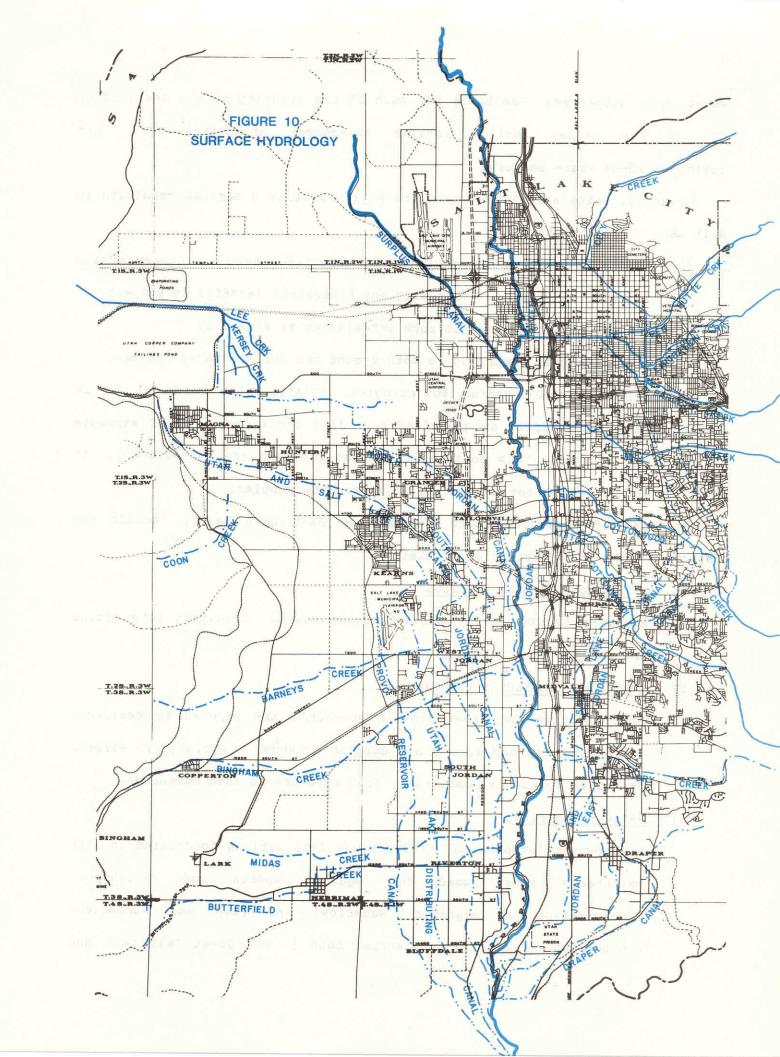
Figure 10 shows the location of major surface hydrological features that directly influence both quantity and quality of available groundwater.

. C. GENERAL LAND USE

Land use is a direct influence on groundwater resources. Urban uses tend to cover permeable ground and increase surface runoff. Agricultural uses allow for groundwater percolation, but also may introduce pollutants from animal waste concentrations and irrigation.

Although urban uses should be expected to decrease the potential for groundwater pollution, they also present hazards. Stormwater detention ponds are a known source of groundwater pollution ¹⁸ and seepage from lawns and gardens in Salt Lake Valley contributes 5% (17,000 acre-feet) of the total annual groundwater recharge. Use of fertilizers on lawns greatly increases potential for contamination from major anions such as phosphate and nitrate.

Perhaps the most serious land use related threat to groundwater resource management is that of landfills or tailing piles containing toxic or hazardous materials. The Resource Conservation Recovery Act set in motion a process to enumerate hazardous wastes and require monitoring of both ground and surface



water where suspected. The basis for much of the priority weight assigned by EPA in the valley shallow aquifer assessment originates from this toxic/hazardous waste concern.

Hazardous waste entrainment into the environment is a serious condition in Salt Lake Valley because:

- 1) Little is known about the types of material perviously dumped or the location of dumping. Virtually any historical landfill is suspect.
- 2) The population exposed to such entrainment is sizeable.
- 3) Entrainment is occurring to both ground and surface water regimes.
- 4) Epidemeological and/or toxicological relationships have yet to be either quantified or qualified regarding local incidence of exposure and effects. This does not mean that problems are non-existent. It means we have not adequately addressed the problem.

Examples of hazardous waste exposure in Salt Lake Valley include the following cases where potential exists:

1) VITRO URANIUM MILL TAILINGS

Exposure to low level radioactive residuals. Transport into ground and surface water exists.

2) ROSE PARK OIL/SLUDGE DISPOSAL

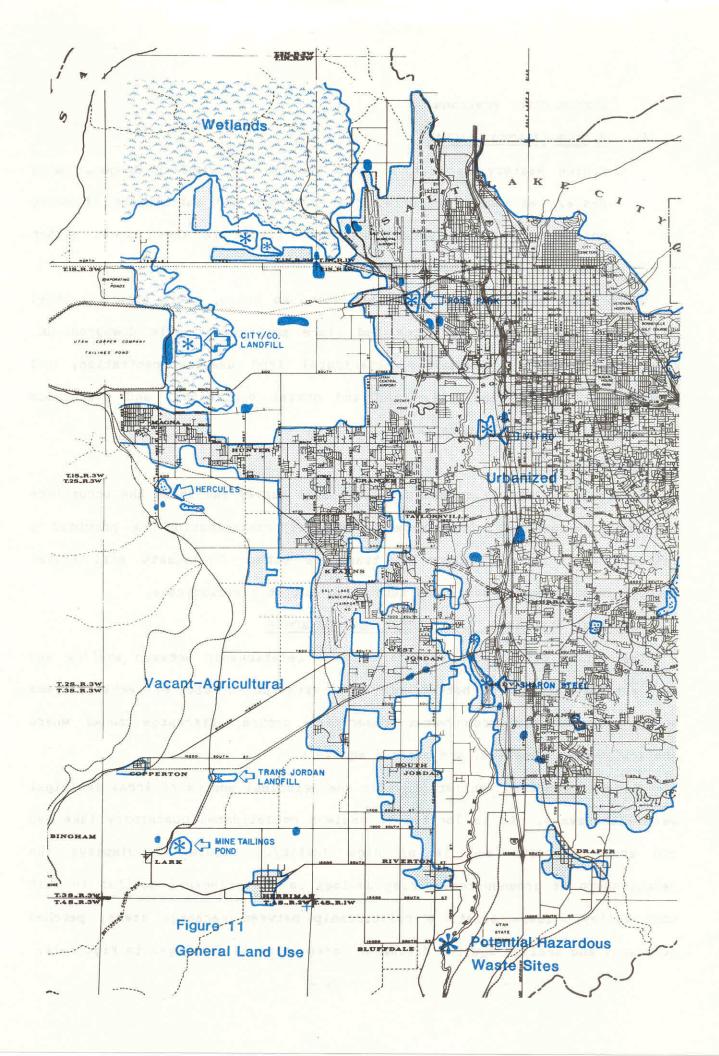
Both volatile and other organic compounds are exposed to residents living on, in, and around old dumping grounds. Carcinogenic effects have been documented nationally from exposure to such compounds.

 \bigcirc

3) VARIOUS LANDFILLS

Groundwater leachate may find its way into wetland foodchains in high watertable zones used for aquatic sports and recreation.

Contamination of migratory waterfowl and fish and subsequent consumption by man is a potential both in the Great Salt Lake and



Jordan River environs.

4) MINING-RELATED LEACHATE

Mining history is rich in both Wasatch and Oquirrh Ranges. Both contain extensive potential for transport and entrainment of heavy metals into public water supplies. Little is known about either potential due to lack of data.

Where these kinds of facilities are known to exist, extra care was taken to assess highest priority sites and place monitoring wells downgradient. Figure 11 generally summarizes urban/rural land use, concentration, and locations of known landfills, sand and gravel operations, and overburden leachate sources in Salt Lake Valley.

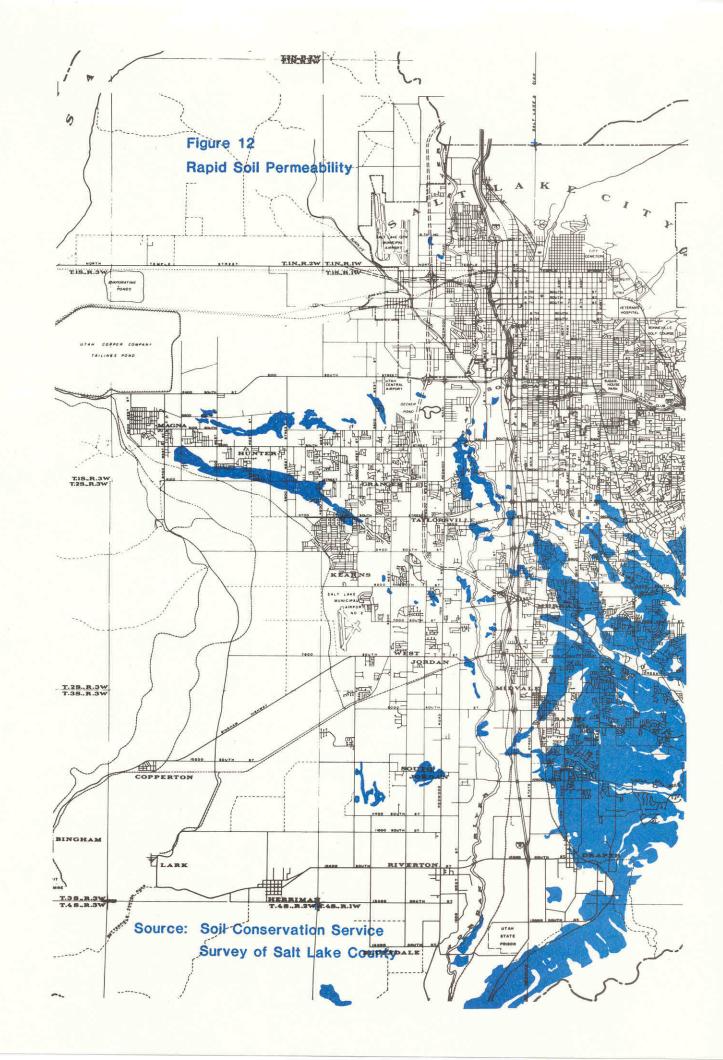
D. SOIL PERMEABILITY

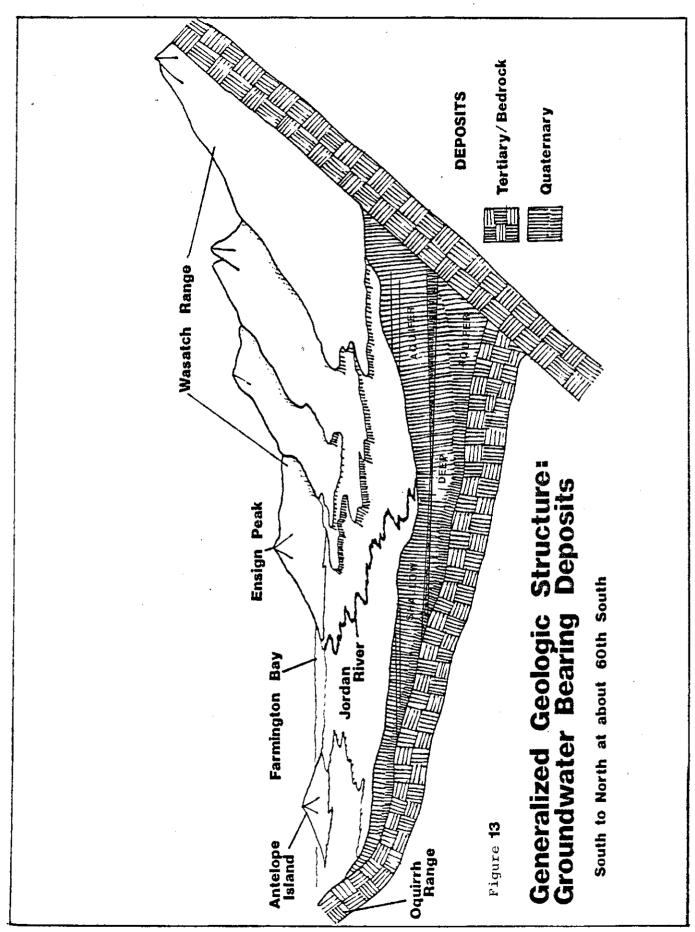
Figure 12 is taken from the Salt Lake Soil Survey and shows the occurrence of highly permeable soils where groundwater contamination is provided a readily available entrainment or transport path. Pollutants will travel through less permeable soils and substrate, but at a slower rate.

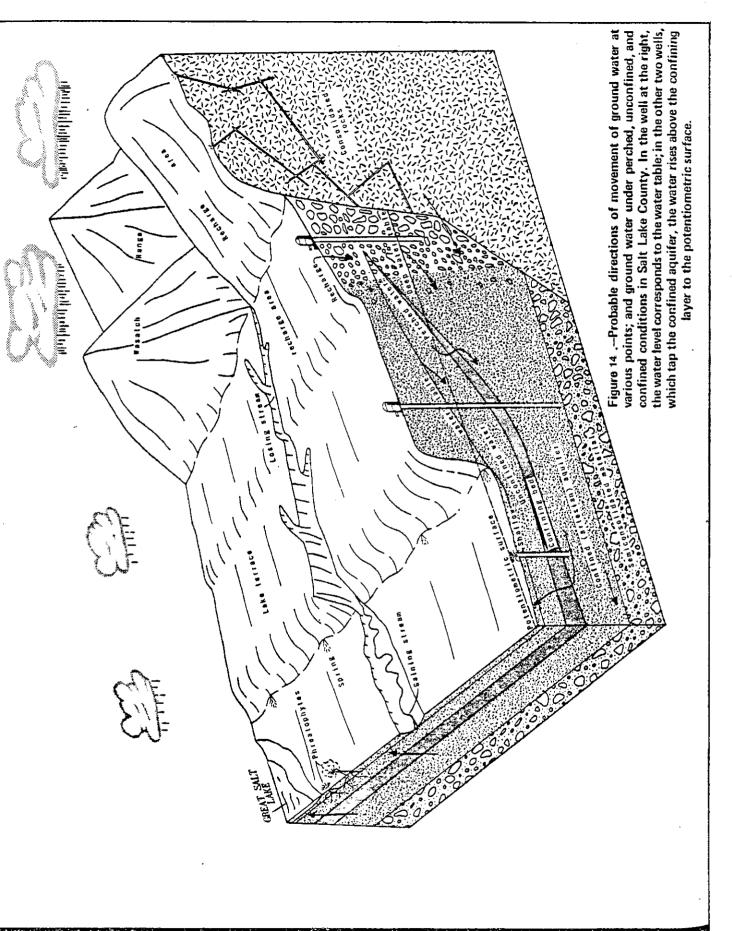
E. RECHARGE, DISCHARGE, AND PERCHED AQUIFER ZONES

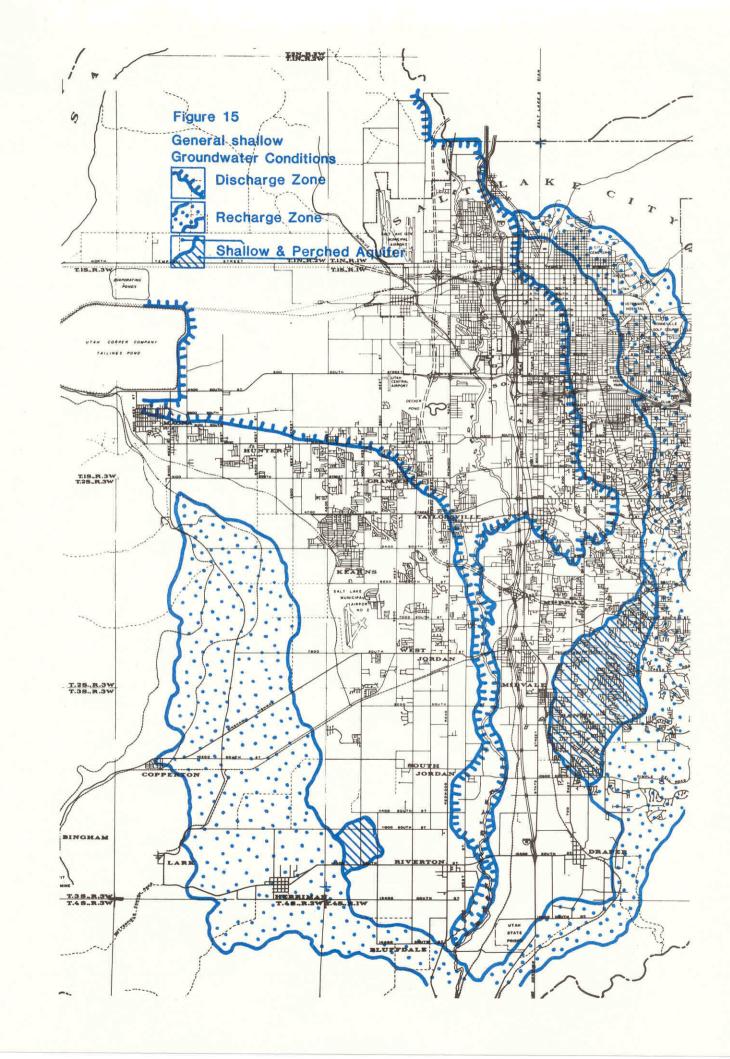
The valley groundwater regimes and the relationship between shallow and deep aquifer resource is better understood with the location of recharge areas where seepage into groundwater reservoirs occurs, discharge zones where shallow aquifers surface, and perched aquifers.

The deep confined aquifer provides the principal source of local municipal well withdrawals. It is located in ancient consolidated quaternary lake bed and shore deposits and is of high quality. Figure 13 displays the relationship of groundwater-bearing geology to the principal aquifer in Salt Lake Valley. There exists a relationship between recharge areas, perched aquifers, and shallow aquifer discharge zones which is displayed in Figure 14.









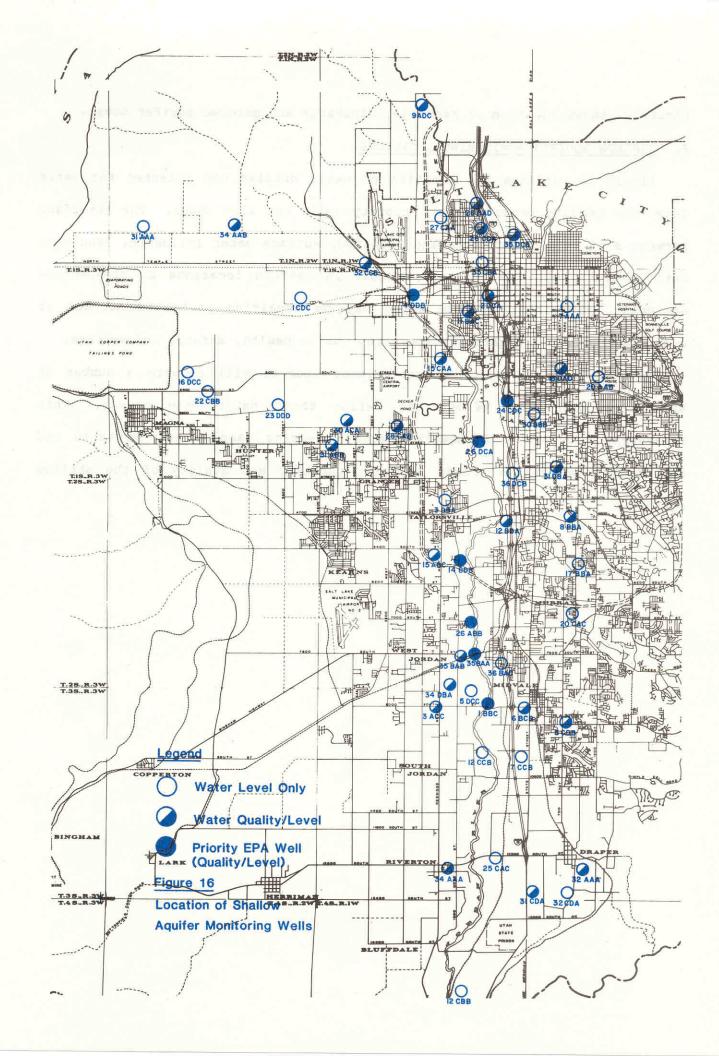


Figure 15 shows location of recharge, discharge and perched aquifer zones.

F. SHALLOW AQUIFER MONITORING LOCATIONS

Figure 16 outlines the location of wells drilled and selected for water level and quality measurement. Priority wells are also shown. The interface between shallow aquifer depth and location, surface water influence. land use, soil permeability, and recharge, result in certain locations as prospective non-point contaminant sources which may be impairing protected surface or subsurface use, and negatively impacting public health, safety, and welfare.

The prospective non-point contaminant sources will require a number of successive monitoring efforts to define their nature and extent. This assessment provides a framework whereby on-going samples may be taken and analyzed to deterimine trends in water quantity and quality of the shallow aquifer.

V. QUALITY OF THE SHALLOW AQUIFER

A. Shallow Groundwater Assessment

As noted before, 55 new wells were drilled and monitored for purposes of quality and quantity assessment. Of the 55 wells, 32 were sampled for quality (7 were EPA designated wells) and all were sampled for quantity (water-level). The quantity of the shallow aquifer has been published in Table 2 of the USGS Report (Appendix). 18

Wells were drilled in 1982 and sampled twice in 1983. Well log data for the 32 quality wells is shown in Figure 17. Water quality data is listed in Table 2. Quality assurance (QA) data, which consists of duplicate, replicate, spikes and blanks, are shown in Table 3.

B. Quality Assurance Assessment and Results.

Both field and laboratory quality assurance plans are required by EPA on groundwater assessments. The U.S. Geological Survey provided a field sampling QA program and the Utah State Medical Laboratory provided the laboratory QA program. Both QA programs were approved by EPA in the final workplan. Such programs are necessary to insure accuracy and integrity of data. During the shallow aquifer assessment, duplicates, spikes, and blanks were processed and submitted for analysis. Examination of QA data reveals the existence of blank sample contamination for TDS, C1, SO4, and Phenols. Investigation of the matter with State Medical Lab personnel indicated the presence of filter contamination. Further details of the nature and extent of contamination are available from open files at the State Medical Laboratory.

STATION

(C-2-1) 12BDA	SAND - D	ry	а	AY -	Brown (Wet)	F	BROWN MUCK	GRAI MDCI	
(C-1-1) 31ABB	CLAY	- Brown (D	ry)		CIAY - B	rown (Wet)	CLAY/FEA GRAVE	L	
(C-1-1) 30ACA	CLAY ~ Brown	(Wet)			CLAY - Gray (W	et)	CLAY - Brown (Dr	y)	
(C-1-1) 28CAB	·								
(C-1-1) 26DCA	SAND -	Dry	SANDY C		CIAY - E	zown (Wet)	CIAY — Gray (W	et)	
(C-1-1)24CDC	CLAY - Black		•		GRAY MUCK		BROWN MUCK		
(C-1-1) 15CAA						**			
(C-1-1) llbac	SANDY CLAY - BI	ack (Dry)			CLAY -	- Gray (Wet)		,	
(C-1-1) 4DDB	CLAY - Brown (Dry)	a	(AY - Brow	n. (We	t)	SANDY	CLAY - Gray (Wet)		
(C-1-1) 2DCA		SAND - 1	Moist			SAND/ PEA GRAVEL	CLAY - Gray (Wet)	1	
(B-1-2) 34AAB	CLAY - Brown (Dry)		LAY - Brow	n (We	t)	SANDY C	LAY - Gray (Wet)	•	
(B-1-1) 35DCB	SANDY	CLAY - Tan				CLAY - Gray (W	et)	· ••	
(B-1-1) 32CCD	CLAY - Brown	(Dry)			BROWN MOCK		GRAY MOCK		
(B-1-1) 26CDA	CLAY - Black		CLAY - Gray (Wet)						
(B-1-1) 26BAD	SANDY CLAY - T	an,	SAND - Gray (Wet)						
(B-1-1) 9ADC	CLAY	- Brown (D	trown (Dry) CLAY - Gray (Wet)						

0-1 1-2 2-3 3-4 4-5 5-6 6-7 7-8 8-9 9-10 10-11 11-12 12-13 13-14 14-15 15-16 16-17 17-18

Well Depth (Ft.)

		·	 ,			····			
(D-3-1) 32AAA	TOP SOIL (DRY)	ry) Sand/fea gravel Sand - Brown (Moist)							
(D-3-1) 31CDA	TOP SOIL (Dry)		cı	ΑY	- Brown (Wet)				
(D-3-1) 6BCB	SAND -	Black	SAND -	Bro	wn (Moist)	CLAY -	- Brown (Wet)		
(D-3-1) 5CDB									
(D-2-1) 8BBA			7.						
(D-1-1) 31DBA	TOP SOI	L (Dry)	YAD	- I	Brown (Dry)	CLAY - 1	Brown (Wet)	CLAY - Gray (Wet)	
(D-1-1) 18DAD	TOP SOIL (Dr	y) a.	AY - Black	CLAY - Brown	(Wet)				
(C-3-1) 34AAA		CIAY - Brown (Wet)							
(C-3-1) 3ACC	TOP SOI	L (Dry)	CLAY	- G1	meen (Wet)	GF	RAVEL		
(C-3-1) lBBC	SAND/PEA CLAY - Br		CLAY - Bla	ack	SANDY CLAY -	Black (Moist)			
(C-2-1) 35BAB	SAND - Dry		CLAY	- B:	rown (Wet)		· · · · · · · · · · · · · · · · · · ·		
(C-2-1) 35BAA	CIAY - Brown (Dry)	CLAY -	Black		CLAY/PI	EA GRAVEL	Sand/Pea (RAVEL	
(C-2-1) 34DDA	SAND - Dry	SANDY CIAY ~	SANDY CL	7X -	Gray (Wet)	a	LAY - Gray (Wet)	
(C-2-1) 26ABB	SAND - Dry			-	SAND/PEA GRAV	FI.			
(C-2-1) 15ABC	CIA	? - Brown (Dry)				EROWN	MUCK		
(C-2-1) 14BDB	SANDY CLAY - Black (Dry)								

0-1 1-2 2-3 3-4 4-5 6-7 7-8 8-9 9-10 10-11 11-12 12-13 13-14 14-15 15-16 15-16 15-16

Mell Depth (Ft.)

TABLE 2. Water Quality Data - Shallow Aguifer Assessment

STATION	STORET	DATE	TIME	TEMP. (Deg. C)	Ph (SU)	COND.	TDS (mg/L)	ALK. (mg/L)	HARD. (mg/L)	Cl (mg/L)	NO3 (mg/L)	NO ₂ (mg/L)	SO, (mg/L)
(B-1-1) 9ADC-1 (B-1-1) 9ADC-1	405007111570901 405007111570901	830201 830801	1100	7.0 15.0	8.0 7.4	6590 8008	4240 4448	1250	170	1300 1650	18.0 7.2	850	240 460
(B-1-1) 26BAD-1 (B-1-1) 26BAD-1	404748111551401 404748111551401	830207 830729	1100	9.0 17.0	7.0 6.5	2530 2500	2150 2730	424	1300	120 94	. 7. 2	<.020	980 1340
(B-1-1) 26CDA-1 (B-1-1) 26CDA-1	404710111551301 404710111551301	830117 830727	1230	11.5 14.5	7.5 7.4	1200 1146	72 4 738	428	560	72 67	2.7 2.18	<.020	150 142
(B-1-1) 32CCD-1 (B-1-1) 32CCD-1	404616111585801 404616111585801	830201 830727	1030	8.0 17.5	7.6 7.2	8590 10611	6432			2900	5.0 .18	<.020	478
(B-1-1) 35DCB-1 (B-1-1) 35DCB-1	404621111550501 404621111550501	830202 830801	1230	11.0 16.0	7.6 7.0	3680 3337	2670 2190	1290	1400	160 190	86.0 36.5	.020	600 510
(B-1-2) 34AAB-1 (B-1-2) 34AAB-1	404702112025201 404702112025201	830201 830726	1230	10.0 17.0	7.8 7.6	14400 825	4502			2500	:1 :32	<.020	138
(C-1-1) 2DCA-1 (C-1-1) 2DCA-1	404527111550102 404527111550102	830107 830728	1045	12.5 17.0	6.9 6.7	2320 2252	1880 1994	373	1400	67 55	13.0 14.57	.030	990 950
(C-1-1) 4DDB-1 (C-1-1) 4DDB-1	404528111570901 ⁻ 404528111570901	830202 830802	1130	7.5 19.5	7.7 7.2	4370 4070	2644			530	:1 :04	<.020	280
(C-1-1) 11BAC-1 (C-1-1) 11BAC-1	404505111552501 404505111552501	830107 830728	1200	10.5 15.0	7.1 7.0	10900 11704	8920 10303	948	3500	1700 4650	<.1 1.84	<.020	3900 4030
(C-1-1) 15CAA-1 (C-1-1) 15CAA-1	404357111562301 404357111562301	821029 830802	1400	15.5	7.2 6.8	3410 2302	1646			135	.06		585
(C-1-1) 24CDC-1 (C-1-1) 24CDC-1	404247111541001 404247111541001	830204 830728	1430	7.0 17.0	7.1 7.0	3220 3006	2432			385	6.0 6.67	<.020	950
(C-1-1) 26DCA-1 (C-1-1) 26DCA-1	404154111545901 404154111545901	830202 830801	1500	8.0 15.0	7-4 7-1	4230 1848	3040 1270	439	1500	650 190	51 5.60	<.020	950 360
(C-1-1) 28CAB-1 (C-1-1) 28CAB-1	404207111574301 404207111574301	830202 830804	1100	13.0 15.0	8.0 7.3	1030 1047	586 588	185	240	170 170	.1	<.020 .26	95 95
(C-1-1) 30ACA-1 (C-1-1) 30ACA-1	404223111592701 404223111592701	830207 830803	1350	9.5 18.0	7.6 7.1	4640 4716	3150 3056	1050	620	540 475 -	8.6 13.9	.020	870 795
(C-1-1) 31A8B-1 (C-1-1) 31ABB-1	404142111594301 404142111594301	830207 830803	1130	10.5 17.5	7.4 6.6	1990 1737	1330 1132	416	580	260 165	5.8 3.5	<.02	320 236
(C-2-1) 12BDA-1 (C-2-1) 12BDA-1	403949111540301 403949111540301	830203 830721	1235	11.0 16.0	7.3 7.1	2310 2229	1340			350	5.5 3.5	<.020	146
(C-2-1) 14BDB-1 (C-2-1) 14BDB-1	403853111552501 403853111552501	830203 830809	1330	9.0 17.0	6.3 6.0	2310 1639	1750 1062	103	900	210 205	.45 1.18	<.020	910 394
(C-2-1) 15ABC-1 (C-2-1) 15ABC-1	403903111561601 403903111561601	830207 830810	1030	10.5 17.0	7.1 6.7	2240 2408	1520 1664	325	800	290 325	5.6 5.04	<.020	480 495
(C-2-1) 26ABB-1 (C-2-1) 26ABB-1	403721111550601 403721111550601	830203 830808	1100	7.0 18.0	7.3 6.8	8960 7963	6138			1125	8.6	<.020	2575
(C-2-1) 34DDA-1 (C-2-1) 34DDA-1	403552111554301 403552111554301	830117 830811	1400	10.0 16.0	7.1 6.7	3450 2494	2480 1666	426	1200	520 330	14	<.020	810 481
(C-2-1) 35BAA-1 (C-2-1) 35BAA-1	403634111551301 403634111551301	830203 830811	1100	9.5 19.0	6.9 6.5	3160 3560	2550 2720	282	1600	300 400	.12	<.02	1200 1215
(C-2-1) 35BAB-1 (C-2-1) 35BAB-1	403632111552301 403632111552301	830117 830811	1300	12.5 16.0	7.0 6.8	2500 3157	1660 2262	332	780	370 410	1.7	.020	450 755

TABLE 2 (cont'd)

STATION	STORET	DATE	TIME	TEMP. (Deg. C)	Ph (SU)	COND.	TDS (mg/L)	ALK. (mg/L)	HARD. (mg/L)	Cl (mg/L)	(mg/L)	NO2 (mg/L)	SO ₄ (mg/L)
(C-3-1) lBBC-1 (C-3-1) lBBC-1	403533111543301 403533111543301	830204 830816	1200	10.5 16.5	6.7 6.3	2340 2730	1520 1766	892	950	250 348	<.1 <.02	<.02	180 266
(C-3-1) 3ACC-1 (C-3-1) 3ACC-1	403517111561301 403517111561301	030114 030816	1400	13.0 19.0	7.2 6.6	2910 4682	1600 3108	455	870	180 145	25.0 238	-54	690 1113
(C-3-1) 34AAA-1 (C-3-1) 34AAA-1	403117111554901 403117111554901	830117 830818	1400	12.5 14.0	7.1 6.7	3470 3538	2810 2809	316	1500	340 345	4.2 4.62	<.02	1300 1290
(D-1-1) 18DAD-1 (D-1-1) 18DAD-1	404348111522201 404348111522201	830107 830805	1115	12.0 17.0	7.0 6.7	1460 1474	778 952	318	570	160 27	7.9 10.6	<.02	220 250
(D-1-1) 31DBA-1 (D-1-1) 31DBA-1	404119111523501 404119111523501	830112 830808	1445	12.0 15.5	7.1 6.5	2040 2558	1270 1618	443	700	350 860	.22	<.02	170 242
(D-2-1) 8BBA-1 (D-2-1) 8BBA-1	404002111520601 404002111520601	830112 830809	1100	11.5 18.0	7.0 6.7	2120 1846	1510 1304	405	920	210 150	.19 .06	<.02	550 430
(D-3-1) 5CDB-1 (D-3-1) 5CDB-1	403501111515501 403501111515501	830204 830817	1100	7.0 20.5	7.6 6.9	394 525	336			34	2.9 4.00	<.02	61
(D-3-1) 6BCB-1 (D-3-1) 6BCB-1	403520111531701 403520111531701	830112 830817	1300	14.5 16.0	7.3 7.0	1560 1848	966 1072	322	500	230 220	4.3 35.40	<u><02</u>	180 195
(D-3-1) 31CDA-1 (D-3-1) 31CDA-1	403038111525501 403038111525501	830112 830818	1130	10.5 15.0	7.3 7.1	2830 2944	1960 1978	266	880	420 405	21 25.00	<.02	530 493
(D-3-1) 32AAA-I (D-3-1) 32AAA-1	403117111511601 403117111511601	830112 830818	1230	11.0 12.0	6.8 6.5	760 1069	490 688	180	250	83 150	4.5 4.82	<.02	100 134

STATION	As (ug/L)	Cđ (ug/L)	Cr (ug/L)	Cu (ug/L)	Fe (ug/L)	Pb (ug/L)	Mn (ug/L)	Eg (ug/L)	Ni (ug/L)	Se (ug/L)	Ag (ug/L)	Ba (ug/L)	Zn (ug/L)	Ca (mg/L)
(B-1-1) 9ADC-1 (B-1-1) 9ADC-1	18	< 1	< 5	< <u>10</u>	540 830	< 5	90 110	<.1	90	1.0		120	 < 5	20
(B-1-1) 26BAD-1 (B-1-1) 26BAD-1	60 95	<1 <1	< 10 < 5	9 25	540 30	1 5	<00 <10	<:1	<10	<.5	 2	< 50	< 5	390
(B-1-1) 26CDA-1 (B-1-1) 26CDA-1	10	~ 1	42	<10	< 3 640	<u></u>	< 10	₹.1	30	<.5	< <u>2</u>	50	< 5	42
(B-1-1) 32CCD-1 (B-1-1) 32CCD-1	300 350	5	11	<10	< 30	< 5	30	₹.1	412	1.5	7	80	< - 5	
(B-1-1) 35DCB-1 (B-1-1) 35DCB-1	14	1	< 5	<10	< 10 < 30	 5	< 10	<.1	<10 ;	<.5	3	50	5	26
(B-1-2) 34AAB-1 (B-1-2) 34AAB-1	3	5	12	<10	< 30	-	155	<.1	250	<.5	5	270	20	
(C-1-1) 2DCA-1 (C-1-1) 2DCA-1	₹. 1	< 1 1	10 < 5	< 10	<30	2 5	510 565	<:1 <:1	13	<.5	2	< 50	< 5	470
(C-1-1) 4DDB-1 (C-1-1) 4DDB-1	130 154	<u><1</u>	< 5	40	< 30	< 5	60	<.1	< 10	<.5	< 2	60	< 5	=
(C-1-1)11BAC-1 (C-1-1)11BAC-1	12.5	<u>-</u>		20	190 <30	20	560 250	<.1	107	1.5	7	< 50	10	310
(C-1-1) 15CAA-1 (C-1-1) 15CAA-1	27	<1	< 5	25	.< 30	~ <u></u>	185	<.1	< 10	<.5	< 2	< 50	<u></u>	=
(C-1-1) 24CDC-1 (C-1-1) 24CDC-1	1.5	< 1	₹ 5	<10	< 30	< 5	10	<.1	< 10	<.5	3	< 50	5	
(C-1-1) 26DCA-1 (C-1-1) 26DCA-1	44 70	<1 <1	10 < 5	< 10 9	30 < 30	< 1 < 5	30 80	<.1 0.7	<10	<.5	2	50	8	170
(C-1-1) 28CAB-1 (C-1-1) 28CAB-1	40 62	< 1 < 1	< 10 < 5	< 1 <10	130 150	< 1 < 5	100 90	<.1 <.1	< 10	<.5	<u></u>	70	< 5	40
(C-1-1)30ACA-1 (C-1-1)30ACA-1	110 95	< 1 < 1	< 10 < 5	10 15	20 30	1 5	100 .90	<:1 <:1	<10	2.5	- <u>-</u>	< 50	15	51
(C-1-1) 31ABB-1 (C-1-1) 31ABB-1	16 18	< 1 < 1	< 5	< 10	< 30 < 30	< ² 5	< 10 < 10	<:1	<10	<.5	< 2	90	<u></u> < 5	90
(C-2-1)12BDA-1 (C-2-1)12BDA-1	2 4.0	1	6			13								
(C-2-1)14BDB-1 (C-2-1)14BDB-1	50 54.0	< 1 < 1	<10 < 5	<u>< 1</u>	37000	<1 <5	480	<.1						180
(C-2-1) 15ABC-1 (C-2-1) 15ABC-1	8 10	< 1 < 1	10 < 5	2	30	<1 <5	<10	<u><.1</u>	=	=				190
(C-2-1) 26ABB-1 (C-2-1) 26ABB-1	60 54	< 1	< 5			< 5	*****		==		=			
(C-2-1) 34DDA-1 (C-2-1) 34DDA-1	6 8	< 1 < 1	10	5	40	< 1 < 5	30	<.1	=	=				260
(C-2-1) 35BAA-1 (C-2-1) 35BAA-1	10 390	200 43	10 5		1900	46 27	1800	<.1					=	390
(C-2-1) 35BAB-1 (C-2-1) 35BAB-1	17 20	< 1 < 1	10 < 5	9	20	< <u>1</u>	30	<.1						170

STATION	As (ug/L)	Cđ (ug/L)	Cr (ug/L)	Cu (ug/L)	Fe (ug/L)	Pb (ug/L)	Mn (ug/L)	Hq (ug/L)	(nd/r)	Se (ug/L)	Ag (ug/L)	Ba (ug/L)	Zn (ug/L)	Ca (mg/L)
(C-3-1) 1BBC-1 (C-3-1) 1BBC-1	51 460	<1	< 10 7	<1	1100	< 1 < 5	510	< <u>.1</u>		=			=	200
(C-3-1) 3ACC-1 (C-3-1) 3ACC-1	99 58	<1	10	. 2	4200	< 1 < 5	670	< <u>.1</u>						230
(C-3-1) 34AAA-1 (C-3-1) 34AAA-1	6.5	₹1	< 5	=	30	< - 5	30			==			=	420
(D-1-1)18DAD-1 (D-1-1)18DAD-1	<.1 <.5	<1 <1	<10 < 5	10	< 3 30	< \frac{1}{5}	< 10 < 3	<:1 <:1	< <u>10</u>	₹.5	< <u>2</u>	₹50	< <u>5</u>	160
(D-1-1) 31DBA-1 (D-1-1) 31DBA-1	2.5	<1	< 5		30	< 5	180				=			160
(D-2-1) 8BBA-1 (D-2-1) 8BBA-1	2.0	<1 <1	<10 < 5	_1	850	< <u>1</u> < 5	530	<u><.1</u>						210
(D-3-1) 5CDB-1 (D-3-1) 5CDB-1	5.5	<1	< 5			<u></u>					=			
(D-3-1) 6BCB-1 (D-3-1) 6BCB-1	14 15	<1 <1	<10 <5	_4	18	14	4	<.1						95
(D-3-1) 31CDA-1 (D-3-1) 31CDA-1	48	<1	₹ 5		40		10							140
(D-3-1) 32AAA-1 (D-3-1) 32AAA-1	1.5	< 1	₹ 5	=	17	< 5	3	==			==		==	62

STATION	Na (mg/L)	K (mg/L)	Mg (mg/L)	P (mg/L)	Silica (mg/L)
(B-1-1) 9ADC-1 (B-1-1) 9ADC-1	1500	54	28	5.3	14
(B-1-1) 26BAD-1 (B-1-1) 26BAD-1	110	77	80	5.3	50
(B-1-1) 26CDA-1 (B-1-1) 26CDA-1	50	7.8	110	1.1	24
(B-1-1) 32CCD-1 (B-1-1) 32CCD-1					
(B-1-1) 35DCB-1 (B-1-1) 35DCB-1	440	42	320	1.1	30
(B-1-2) 34AAB-1 (B-1-2) 34AAB-1		740 770			Ξ
(C-1-1) 2DCA-1 (C-1-1) 2DCA-1	60	8.1	50	<u>.7</u>	19
(C-1-1) 4DDB-1 (C-1-1) 4DDB-1					==
(C-1-1) 11BAC-1 (C-1-1) 11BAC-1	1600	200	660	-6	31
(C-1-1) 15CAA-1 (C-1-1) 15CAA-1					
(C-1-1) 24CDC-1 (C-1-1) 24CDC-1					
(C-1-1) 26DCA-1 (C-1-1) 26DCA-1	430	81	250	1.2	31
(C-I-1) 28CAB-1 (C-I-1) 28CAB-1	120	20	34	.7	24
(C-1-1) 30ACA-1 (C-1-1) 30ACA-1	920	44	120	2.6	32
(C-1-1) 31ABB-1 (C-1-1) 31ABB-1	230	37	86	<u>.7</u>	39
(C-2-1)12BDA-1 (C-2-1)12BDA-1					
(C-2-1)14BDB-1 (C-2-1)14BDB-1	180	15	110	3.1	27
(C-2-1) 15ABC-1 (C-2-1) 15ABC-1	180	25	80	.9	32
(C-2-1) 26ABB-1 (C-2-1) 26ABB-1			==		==
(C-2-1) 34DDA-1 (C-2-1) 34DDA-1	340	17	140	.5	33
(C-2-1) 35BAA-1 (C-2-1) 35BAA-1	190	17	140	2.9	17
(C-2-1) 35BAB-1 (C-2-1) 35BAB-1	250	12	87	.6	43

STATION	Na (mg/L)	(mg/L)	Mg (mg/L)	P (mg/L)	Silica mg/L)
(C-3-1) 1BBC-1 (C-3-1) 1BBC-1	200	32	110	.9	39
(C-3-1) 3ACC-1 (C-3-1) 3ACC-1	140	36	72	.3	30
(C-3-1)34AAA-1 (C-3-1)34AAA-1	280	33	98	.7	<u>37</u>
(D-1-1) 18DAD-1 (D-1-1) 18DAD-1	100	4.5	41	.2	14
(D-1-1) 31DBA-1 (D-1-1) 31DBA-1	180	13	73	6	22
(D-2-1) 8BBA-1 (D-2-1) 8BBA-1	140	9.9	96	-4	21
(D-3-1) 5CDB-1 (D-3-1) 5CDB-1	==				
(D-3-1) 6BCB-1 (D-3-1) 6BCB-1	150	14	. 63	.8	27
(D-3-1) 31CDA-1 (D-3-1) 31CDA-1	270	25	130	1.3	49
(D-3-1) 32AAA-1 (D-3-1) 32AAA-1	66	3.4	22	<.1	13

STATION	Methyluromide, Total (ug/L)	Methylenechloride, Tota: (ugʻii	Bromoform, Total (ug/L)	Chloroform, Total (ug/L)	Chlorodibromomethane, Total, (ug/L)	Dichlorobromomethane, Total (ug/L)	Dichlorodifluro-methane, Total (ug/L)	Trichlorofluoromethane, Total, (ug/L)	Carbontetrachloride, Total, (ug/L)	Chloroethane, Total (ug/L)	1,1-Dichioroethame, Total, {ug/l}	1,2-Dichloroethane, Total (ug/L)	1,1,1-Trichloroethane Total, (ug/L)	1,1,2-Trichloroethane, Total (ug/L)
(B-1-1) 26BAD-1 (B-1-1) 26BAD-1	<1.0	<1.0	< 1.0	<1.0	<1.0	<1.0	< 1.0	<1.0	< 1.0	<1.0	<1.0	<1.0	<1.0	<1.0
(B-1-1) 32CCD-1 (B-1-1) 32CCD-1	< 1.0	<1.0	<1.0	<1.0	< 1.0	<1.0	< 1.0	< 1.0	<1.0	<1.0	<1.0	<1.0	< 1.0 .01	< 1.0
(B-1-2) 34AAB-1 (B-1-2) 34AAB-1	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	< 1.0	<1.0	<1.0	<1.0	<1.0	< 1.0 .01	<1.0 <1
(C-1-1) 4DDB-1 (C-1-1) 4DDB-1	<1	<1	<u><1</u>	<1	<1	<u><1</u>	< <u>1</u>	<1	< 1 < 1	<1	<1 <1	<1 <1	<1 .008	< 1 < 1
(C-1-1) 24CDC-1 (C-1-1) 24CDC-1	<1.0	<1.0	<1.0	<1.0	<1.0	<u><1.0</u>	<pre>< 1:0</pre>	< 1.0	<1.0 <1.0	<1.0	20 <1.0	<1.0 <1.0	< 1.0 2.5	<1.0 <1.0
(C-1-1) 26DCA-1 (C-1-1) 26DCA-1	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	< 1.0 < 1.0	<1.0	<1.0 <.10	<1.0	<1.0 <1.0	< 1.0 < 1.0	<1.0	<1.0 <1.0
(C-1-1) 28CAB-1 (C-1-1) 28CAB-1	<1.0	<1.0	<1.0	<1.0	< <u>1.0</u>	<1.0	< 1.0 1.0	< 1.0	< 1.0 < 1.0	<1.0	<1.0 <1.0	< 1.0 < 1.0	< 1.0	<1.0 <1.0
(C-1-1) 30ACA-1 (C-1-1) 30ACA-1	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	< 1.0 < 1.0	<1.0	<1.0 .004	<1.0	<1.0 <1.0	< 1.0 < 1.0	< 1.0	< 1.0 < 1.0
(C-1-1) 31ABB-1 (C-1-1) 31ABB-1	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0 <1.0	<1.0	<1.0 .005	<1.0	<1:0 <1:0	<1.0 <1.0	<1.0 .013	<1.0 <1.0
(C-2-1) 12BDA-1 (C-2-1) 12BDA-1	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0 <1.0	<1.0	<1.0 <1.0	<1.0	<1.0 <1.0	<1.0 <1.0	<1.0 .01	<1.0 <1.0
(C-2-1) 14BDB-1 (C-2-1) 14BDB-1	< 1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0 <1.0	<1.0	<1.0 .003	<1.0	<1.0 <1.0	<1.0 <1.0	<1.0 <1.0	<1.0 <1.0
(C-2-1) 15ABC-1 (C-2-1) 15ABC-1	<1.0	<1.0	<1.0	8.0	<1.0	<1.0	<1:0	<1.0	<1.0	<1.0	<1.0 <1.0	<1.0 <1.0	<1.0 <1.0	< 1.0 < 1.0
(C-2-1) 26ABB-1 (C-2-1) 26ABB-1	< 1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0 <1.0	<1.0	<1.0	<1.0	<1.0 <1.0	<1.0 <1.0	<1.0 <1.0	<1.0 <1.0
(C-2-1) 35BAA-1 (C-2-1) 35BAA-1	<1.0	<1.0	<1.0	2.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0 <1	<1.0 .006	<1.0 <1
(C-3-1) 1BBC-1 (C-3-1) 1BBC-1	< 1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0 <1	<1.0	<1.0	<1.0	<1.0	11 <1.	<1.0 <1	<1.0
(D-3-1) 5CDB-1 (D-3-1) 5CDB-1	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0	<1.0 <1	<1.0	<1.0	<1.0	<1.0 <1	<1.0 <1	<1.0	<1.0 <1

STATION	1,1,2,2-Tetrachloro- ethane, Total (ug/L)	Chloroethylene, Total (Uj/L)	l,l-Dichloroethylene, Total, (ug/L)	Trichloroethylene, Total (ug/L)	Tetrachloroethylene, Total (ug/L)	1,2-Dichloropropane, Total (ug/L)	1,3-Dichloropropane, Total (ug/L)	Ethyl Vinyl Ether, Total (ug/L)	Benzene, Total (ug/L)	Chlorobenzene, Total (ug/L)	Toluene, Total (ug/L)	Ethylbenzene, Total (ug/L)	Dichlorodibromo- methane, Total (ug/L)	Phenols, Total (ug/L)
(B-1-1) 26BAD-1 (B-1-1) 26BAD-1	<1.0	<1.0 <999*	<1.0	<1.0	<1.0 .006	<1.0	<1.0	<1.0	< 2.0 < 999*	<1.0 <999*	<1.0 <999*	<1.0 <999*	.3	< 1
(B-1-1) 32CCD-1 (B-1-1) 32CCD-1	<1.0	< 1.0 < 999*	<1.0	<1.0 <1	<1.0 .13	<1.0	<1.0	<1.0	<1.0 <999*	<1.0 <999*	<1.0 <999*	<1.0 <999*	.01	3 2
(B-1-2) 34AAB-1 (B-1-2) 34AAB-1	<1.0	<1.0 <999*	<1.0	<1.0	<1.0 .07	<1.0	<1.0	<1.0	<1.0 <999*	<1.0 <999*	<1.0 <999*	<1.0 <999*	<1	35 < 1
(C-1-1) 4DDB-1 (C-1-1) 4DDB-1	<1 <1	<1 <999*	<1 <1	<1 <1	<1 .003	<1	<u><1</u>	<1	<1 <999*	<1 <999*	<1 <999*	<1 <999*	.003	190 1
(C-1-1) 24CDC-1 (C-1-1) 24CDC-1	<1.0 <1.0	11 < 999*	<1.0 <1.0	8.0 4.2	3.0 2.7	<u><1.0</u>	<1.0	<1.0	< 2.0 < 999*	<1.0 <999*	<1.0 <999*	<1.0 <999*	<1	33 2
(C-1-1) 26DCA-1 (C-1-1) 26DCA-1	<1.0 <1.0	< 1.0 < 999*	<1.0 <1.0	<1.0 <1.0	<1.0 .30	<1.0	<1.0	<1.0	< 1.0 < 999*	<1.0 <999*	<1.0 <999*	<1.0 <999*	.08	32 <1
(C-1-1) 28CAB-1 (C-1-1) 28CAB-1	<1.0 <1.0	<1.0 <999*	<1.0 <1.0	<1.0 <1.0	<1.0 .01	<1.0	<1.0	<1.0	<1.0 <999*	<1.0 <999*	<1.0 <999*	<1.0 <999*	.018	< 1
(C-1-1) 30ACA-1 (C-1-1) 30ACA-1	<1.0 <1.0	< 1.0 < 999*	<1.0 <1.0	<1.0 <1.0	<1.0 .066	<1.0	<1.0	<1.0	< 1.0 < 999*	<1.0 <999*	<1.0 <999*	<1.0 <999*	.002	< 1
(C-1-1) 31ABB-1 (C-1-1) 31ABB-1	<1.0 <1.0	< 1.0 < 999*	<1.0 <1.0	<1.0 <1.0	<1.0 .045	<1.0	<1.0	<1.0	<1.0 <999*	<1.0 <999*	< 1.0 < 999*	< 1.0 < 999*	.014	< 1
(C-2-1) 12BDA-1 (C-2-1) 12BDA-1	< 1.0 < 1.0	< 1.0 < 999*	< 1.0 < 1.0	<1.0 <1.0	<1.0 .01	<1.0	<1.0	<1.0	< 1.0 < 999*	< 1.0 < 999*	< 1.0 < 999*	< 1.0 < 999*	<1	4 5
(C-2-1)148DB-1 (C-2-1)148DB-1	<1.0 <1.0	< 1.0 < 999*	< 1.0 < 1.0	< 1.0 < 1.0	<1.0 <1.0	<1.0	<1.0	<1.0	< 1.0 < 999*	< 1.0 < 999*	<1.0 <999*	< 1.0 < 999*	<1	660 1
(C-2-1)15ABC-1 (C-2-1)15ABC-1	< 1.0 < 1.0	< 1.0 < 999*	< 1.0 < 1.0	<1.0 <1.0	< 1.0	<1.0	<1.0	<1.0	<1.0 <999*	< 1.0 < 999*	< 1.0 < 999*	< 1.0 < 999*	<1	< 1
(C-2-1) 26ABB-1 (C-2-1) 26ABB-1	< 1.0 < 1.0	< 1.0 < 999*	< 1.0 < 1.0	< 1.0 < 1.0	<1.0 .018	<1.0	<1.0	<1.0	< 1.0 < 999*	<1.0 <999*	< 1.0 < 999*	< 1.0 < 999*	<1	10 < 1
(C-2-1) 35BAA-1 (C-2-1) 35BAA-1	<1.0	< 1.0 < 999*	<1.0	< 1.0	<1.0	<1.0	<1.0	<1.0	< 1.0 < 999*	< 1.0 < 999*	< 1.0 < 999*	< 1.0 < 999*	<1	<1
(C-3-1) lBBC-1 (C-3-1) lBBC-1	<1.0	< 1.0 < 999*	<1.0	< 1.0	<1.0	<1.0	<1.0	<1.0	400 < 999*	<1.0 <999*	8.0 < 999*	< 1.0 < 999*	<1	44
(D-3-1) 5CDB-1 (D-3-1) 5CDB-1	<1.0	< 1.0 < 999*	< 1.0	<1.0	<1.0 .7	<1.0	<1.0	<1.0	< 1.0 < 999*	< 1.0 < 999*	< 1.0 < 999*	< 1.0 < 999*	.2	

^{* =} Minimum Detection Limit is lppm(1000 ug/L)

TABLE 3. Quality Assurance Data - Shallow Aquifer Assessment

STATION	ST	ORET	DATE	TOS (mg/L)	Cl (mg/L)	NO3 (mg/L)	SO ₄ (mg/L)	As (ug/L)	Cđ (ug/L)	Cr (ug/L)	Pb (ug/L)				
(C-1-1) 26DCA-1 DUPLICATE		111545901 111545901		1270 1244	190 170	5.60 4.90	360 350	70 67	<1 <1	< 5 < 5	< 5 < 5				
(C-2-1) 26ABB-1 DUPLICATE	403721 403721	111550601 111550601	830808 830808	6138 6004	1125 1150	.40 .40	2575 2520	54 64	<1 <1	< 5 < 5	< 5 < 5				
(D-3-1) 5CDB-1 DUPLICATE	4035011 4035011	111515501 111515501	830817 830817	336 322	34 34	4.00 4.10	61 52	5.5 5.2	<1 <1	< 5 < 5	< 5 < 5				
OC SPIKE SPIKE RESULTS			830816 830909	66 96	20.5 25	1.32 1.40	12.0 19	22 30	2,9	11 15	22 25				
FIELD BLANK (FI UNFILTERED BLAN	ltered) K		830808 830808	184 < 5	34 <1	.04	114 <10	<.5 <.5	<1 <1	< 5 < 5	< 5 < 5				
STATION	VOA7 (ug/L)	VOA9 (ug/L)	VOAll (ug/L)	VOAL2 (ug/L)	VOAL3 (ug/L)	VOA14 (ug/L)	VOA15 (ug/L)	VOA16 (ug/L)	VOA17 (ug/L)	VOA18 (ug/L)	VOA19 (ug/L)	VOA23 (ug/L)	VOA24 (ug/L)	VOA25 (ug/L)	
(C-1-1)26DCA-1 DUPLICATE	<1 <1	.10	<1 <1	< 1 < 1	.15	<1 <1	<1 <1	< 999* < 999*	<1 <1	<1 <1	.30	< 999* < 999*	< 999* < 999*	< 999* < 999*	
(C-2-1) 26ABB-1 DUPLICATE	<1 <1	.003	<1 <1	<1 <1	<1 <1	<1 <1	< 1 < 1	< 999* < 999*	< 1 < 1	< 1 < 1	.018 .025	< 999* < 999*	< 999* < 999*	< 999* < 999*	
(D-3-1) 5CDB-1 DUPLICATE	< 1 < 1	. 2	<1 1	<1 <1	:4	<1 <1	<1 <1	< 999* < 999*	< 1 < 1	<1 <1	1:0	< 999* < 999*	< 999* < 999*	< 999* < 999*	
FIELD BLANK (Filt UNFILT. BLANK	:-)<1 <1	.003	<1 <1	₹ 1	< <u>1</u>	< 1 < 1	< 1 < 1	< 999* < 999*	< 1 < 1	< 1 < 1	.003 .003	< 999* < 999*	< 999* < 999*	< 999* < 999*	
STATION	VOA26 (ug/L)	VOA27 (ug/L)	Phenols (ug/L)			•						_			
(C-1-1) 26DCA-1 DUPLICATE	<999* <999*	.08	<1 <1		Г		· ·	 -	م ا	genc		·			
(C-2-1)26ABB-1 DUPLICATE	<999* <999*	<1 .001	<u> </u>						LC	9 e iic	ı				
(D-3-1) 5CDB-1 DUPLICATE	<999* <999*	.2	2 <1			VOA7	meth	lorodiflur ane, Total	-	V	CA17		ichloroeth , (ug/L)	ylene,	
FIELD BLANK(Fil-		<1 <1	5 5			eacy.		ontetrachl	oride,	V	CA18	Trich (ug/L)	loroethyle)	ne, Total	
OUTIEI: BERNY	1993-	~*	3			VOA.11	1,1-	l, (ug/L) Dichloroet	hane,		DAI9		hloroethy. (ug/L)	lene,	
t s Minimum Pak	antin- •	(=15 1- ·	1/10=	 .	ļ	VOA12		l, (ug/L)	hana <i>ma</i> l-		OA23	Benzer (ug/L)	me, Total		
* = Minimum Deta	ection T	AMIT 18	τ Ъ ΣΕ (1000	ug/L)		ATRIT.	1,2-1 (ug/1	Dichloroet L)	mone, 100		OA24	-	benzene, 1	(otal	
						VOA13		l-Trichlor 1, (ug/L)	oethane			(ug/L)			
						VOA14	1,1,	2-Trichlor	oethane, 1		0A25	Toluer (ug/L)	e, Total		
						(ug/L) VOR26 Ethylbenzene, Tot				tal					

VOA15

VOA16

1,1,2,2-Tetrachloroethane, Total (ug/L)

Chloroethylene, Total (ug/L)

(ug/L)

VOA27

Dichlorodibromomethane, Total (ug/L)

General Discussion.

Given two samples, taken at a 4 to 6 month interval, the following general observations can be made, together with some comparisons of up and down-gradient data. Given a larger data base, the observations can be verified or rejected but such data is not yet available:

1) The pH of the shallow aquifer ranges between 6.0 and 8.0 units. None of the wells exhibited a pH lower than 6.0 or greater than 8.0. The lowest observed mean pH was 6.1 units. A brief examination of tables of chemical analyses of groundwater in the principal aquifer in Salt Lake County (U.S. Geological Survey, April 1984) reveals only four ph values less than 7.0 and only one less than 6.9. Table 2 of the report shows that 18 of the 32 wells in the shallow aquifer had measured ph's less than 7.0. This may indicate widespread contamination of the shallow aquifer by either organic or mineral acids. Further extensive monitoring of the shallow aquifer is necessary to corroborate such contamination.

mineralencountered at specific wells located near the Jordan River.

- 2) High specific conductivity, Total Dissolved Solids, (TDS), hardness and associated cation concentrations were observed primarily at wells located in the northern quadrants reflecting highly mineralized quality of the shallow aquifers found in these areas. Other areas reflecting higher than average specific conductivity, TDS, hardness and associated cation levels were also encountered at specific wells located near the Jordan River.
- 3) Nitrate levels (mean greater than 10mg/1) were encountered in the northern quadrants of the County in addition to locations along the eastern fringes of the valley and near the prison (southwestern quadrant). These

elevated levels were found located in conjunction with suspected problem land use areas.

- 4) The only metals that appear to be detectable at higher than background levels are Arsenic (As), Iron (Fe) and Manganese (Mn). High arsenic and iron levels occur near the airport and near the Jordan River at about 7800-9000 South Street. High manganese levels occur in shallow wells in the downtown Salt Lake City area and in the areas of high iron and arsenic near 7800-9000 South Streets.
- 5) Elevated levels of organic parameters (27 volatile organics and phenols) do occur in some selected shallow aquifer observation wells, (such as 14BDB-1, and 1BBC-1) but the extent of adverse public health implications cannot presently be defined. Detailed site-specific investigation of these sites is necessary to further determine the extent of the problem.
- Nine (9) sets of shallow observation wells were installed above and/or below specific land uses that were suspected to or at least have the potential to contaminate the shallow and/or deep aquifer(s). The increases (or decreases) of constituent levels is discussed below on a site-by-site basis.
- 1) Well (B-1-2)34AAB-1 is located downgradient from the inactive Salt Lake City landfill. The quality of the shallow aquifer at this location indicates high specific conductance and chloride (C1), which could be due to location in a groundwater discharge area, and the highest barium (Ba) level of any sample.
- 2) Wells (C-1-1)4DDB-1 and (B-1-1)32CCD-1 are located up-and downgradient respectively, from an inactive landfill at approximately North Temple and 4000 West Street. Specific conductance, TDS, C1-, NO₃, SO₄, (As) chromium (Cr), nickel (Ni) and silver (Ag) concentrations increased by approximately 2x, 2x, 5x, 10x, 2x, 2x, 40x and 3x, respectively, while copper (Cu) and

concentrations decreased by approximately 1/4 and 1/2 respectively. The level of Ni in the downgradient well was the highest recorded during the sampling period.

- 3) Wells (D-1-1)31DBA-1 and (C-1-1)26DCA-1 are located up-and downgradient from an inactive landfill located at approximately 3900 South 100 West Streets. Concentrations of hardness, nitrate (NO_3) sulphate (SO_4) , As, sodium (Na), potassium (K), magnesium (Mg) and flouride (F) incresed by approximately 2x, 10x, 4x, 20x, 3x, 3x, 3x and 2x, respectively, while concentrations of Mn decreased by about 1/2.
- 4) Wells (D-3-1)5CDB-1 and (D-3-1)6BCB-1 are located up-and downgradient from an inactive landfill at approximately 9000 South and 300 East Streets. Concentrations of specific conductance, Cl, NO_3 . SO_4 and As increased by approximately 4x, 7x, 4x, 3x and 3x, respectively.
- 5) Wells (C-1-1)24CDC-1 and (C-1-1)26 DCA-1 are located upgradient from the inactive Vitro tailings site. The data can be used in conjunction with specific vitro site data discussed later.
- 6) Well (C-3-1)1BBC-1 and wells (C-2-1)26ABB-1 and (C-2-1)35BAA-1 are located up-and downgradient from the inactive Sharon Steel site (approximately 8000-7800 South 700 West Streets). Specific conductance, TDS, hardness, Cl, NO₃ SO₄, cadmium (Cd), Fe, lead (Pb), Mn and F increased by approximately 3x, 2x, 2x, 2x, 8x, 10x, 2x, 10x, 3x and 3x, respectively, while K concentrations decreased by about 1/2. However, the levels are quite high when compared with other monitoring sites located throughout the valley.
- 7) Wells (B-1-1)35DCB-1 and wells (B-1-1)26BAD-1 and (B-1-1)26 CDA-1 are located up-and downgradient from the Rose Park uncontrolled hazardous waste disposal site and in the vicinity of the AMOCO Oil Refinery (discussed

later). The Rose Park site has been extensively studied by US EPA and will therefore not be discussed here.

In summary, it appears as though leachate from old (inactive) landfills is entering the shallow groundwater system. This leachate, along with uncontaminated groundwater, is not being used for production purposes (domestic or industrial) but may be used as irrigation water where it enters irrigation systems in areas of groundwater discharge. Potential for impairment of stock animals and consumption by man may exist. Leachate is also probably entering surface water systems through the same mechanism, particularly in areas of upward artesian pressure where surface discharge of groundwater occurs.

Bioassay of benthic invertebrates and higher foodchain organisms, such as waterfowl or fish consumed by man, is adviseable to determine potential wildlife use or human health impairment.

Shallow aquifer quality monitoring should be continued by either health or water quality planning agencies at least on a quarterly (if not more frequent) basis, to provide data to more fully understand the hydrologic processes taking place.

C. SITE SPECIFIC INVESTIGATIONS AND DATA

1. AMOCO Oil Company - Salt Lake City Refinery

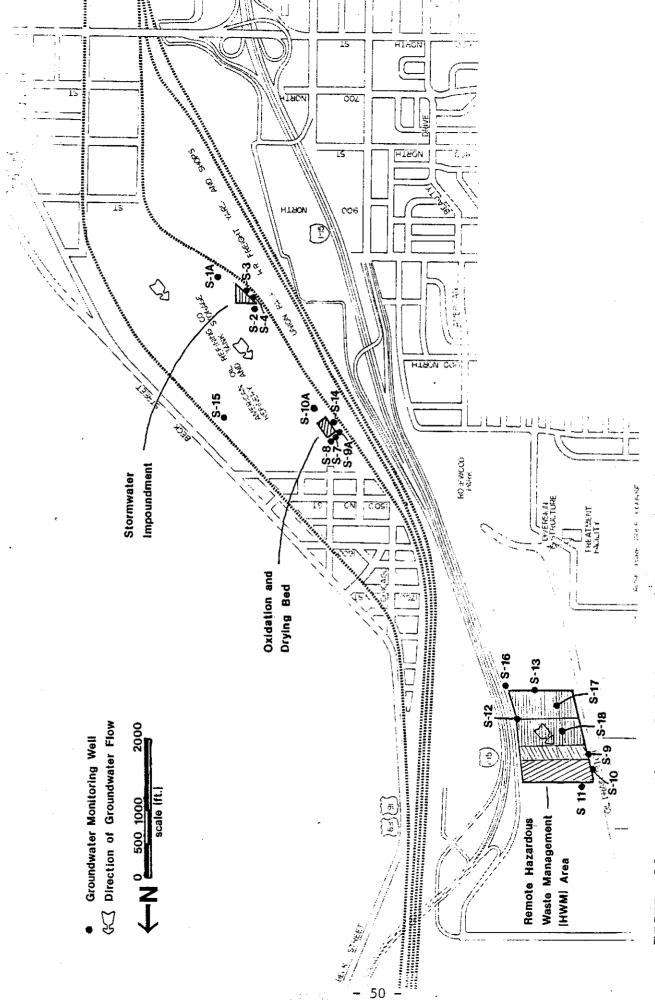
Shallow groundwater quality monitoring at the AMOCO Oil Company - Salt Lake City Refinery was initiated in 1981 by AMOCO. Wells were drilled and are sampled by the company. Samples are analyzed by a private laboratory. Results are sent to the State Department of Health, Bureau of Solid and Hazardous Waste. The wells are located on AMOCO property up and downgradient from three company facilities: stormwater impoundment, remote hazardous waste management area, and an oxidation and drying bed. The location of the

wells are shown in Figures 18.

physical well data (date installed, groundwater and surface elevations, etc.) are listed in Table 4. Note that groundwater elevations in March, 1983 for wells S-10A, S-7, S-8 and S-9A indicate that the groundwater elevation for the upgradient well (S-10A) is lower than the groundwater elevation for the downgradient wells (S-7, S-8 and S-9A). This has been noted by AMOCO and reported to the State. Upgradient well S-15 has a higher groundwater elevation than downgradient well S-14 which corresponds with elevations in adjacent wells (S-7, S-8 and S-9A). The discrepency water level in well S-10A could be due to localized depression in groundwater surface because the overall direction of groundwater movement in the area is from well S-15 towards S-7, S-8, S-9A, and S-14 (east to west). However, groundwater quality, discussed below, shows some discrepancy with this generalization.

Groundwater quality from the AMOCO monitoring wells is shown in Table 5. Quality data near the hazardous waste management area (upgradient wells S-12, S-13 and S-16 and downgradient wells S-9, S-10, S-11, S-17 and S-18) indicate no substantial increase in pH, NO₃, SO₄, Ag, As, Cd, Cr, Hg, Pb, Se, F. All radiologics, pesticides, herbicides, phenol, and total coliform levels were quite high. There is a substantial increase in temperature (about 10C), specific conductance (50%), Na (50%), Ci-(75%), total organic carbon (TOC) (15%) and total organic halide (TOX) (600%) levels. Background levels of specific conductance and TOC were high. Levels of Ba decreased about 50%.

Quality data near the stormwater impoundment (upgradient well S-1A) and downgradient wells (S-2, S-3 and S-4) indicate a substantial increase in specific conductance (20%), SO_4 (50%), C1-(300%), F (30%), TOC (100), TOX (300%) and phenol (20%) levels with high background levels of these



Location of Shallow Aquifer Monitoring Wells; AMOCO Oil Company - Salt Lake City Refinery FIGURE 18.

TABLE 4. Physical Well Data; AMOCO Oil Company - Salt Lake City Refinery

AMOCO WELL DATA

Well No.	Relative Location	Date Installed	Ground Elevation (ft. MSL)			Groundwater Elevation March 14-18, 1983 (ft. MSL)
			-	101	200	(222 122)
S-lA	UPGRADIENT	9/24/81	4225.00	4222.00	4208.00	4219.71
S-2	DNGRADIENT	9/11/81	4225.84	4220.84	4208.84	4218.49
S-3	DNGRADIENT	9/11/81	4225.32	4220.32	4208.32	4219.75
S-4	DNGRADIENT	9/11/81	4225.45	4220.95	4208.45	4218.22
ş						
S-10A	UPGRADIENT	9/25/81	4218.17	4214.17	4200.17	4216.87
S-7	DNGRADIENT	9/24/81	4218.34	4217.34	4202.34	4216.96
5-8	DNGRADIENT	9/24/81	4218.69	4216.19	4202.19	4216.98
S-9A	DNCRADIENT	9/24/81	4218.62	4215.62	4201.62	4217.10
				•		
S-12	UPCRADIENT	9/10/81	4213.68	4209.68	4194.68	4211.14
S-13	UPGRADIENI	9/25/81	4212.79	4209.79	4195.79	4210.61
		,,				
S-9	DNCRADIENT	9/10/81	4210.78	4207.78	4190.78	4208.77
S-10	DNGRADIENT	9/10/81	4211.79	4209.79	4194.79	4209.07
S-11	DNGRADIENT	9/10/81	4210.29	4208.29	4193.29	4209.52
S-15	UPGRADIENT	3/9/83	4223.43	4209.39	4204.39	4220.77
S-14	DNGRADIENT	3/9/83	4224.29	4205•27	4203.27	4217.09
S-16	UPGRADIENT	3/10/83	4217.71	4202.45	4197.45	4212.16
S-17	DNGRADIENT	3/10/83	421.3.53	4198.37	4193.37	4210.96
S-18	DNGRADIENT	3/10/83	4214.27	4199.07	4194.07	4211.04

Note: All well diameters are 2.0 inches. All screen slot sizes are 0.010 inches.

TABLE 5. Shallow Groundwater Quality; AMOCO Oil Company - Salt Lake City Refinery

Well two.	Date (yymmdd)	Temp	pH (su)	Specific Conduct. (uMHO/cm)	NO3-N (mg/L)	SO ₄	Total Coliform (MPN/100ml)	Relative Location
S-la S-la S-la S-la	811119 820316 820622 820921	16.7 15 21 17.8	7.42 7.51 6.70 6.45	6200 5600 12000 4550	.04 .16 <.01 .39	93 1680 540 420	330 <2 <2 <2	UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT
\$-2 \$-2 \$-2 \$-3 \$-3 \$-3 \$-3 \$-4 \$-4 \$-5	811119 820316 820622 820921 811119 820316 820622 820921 811119 820316 820622 820921	18.6 21 19.4 18.6 16 19 20.5 19.1 15 18 20.5	6.72 6.79 6.59 6.27 6.70 6.54 6.35 6.14 6.59 6.30 6.33	9400 12400 11000 5200 6400 11500 5800 4650 7700 11200 7700 5000	.08 <.01 <.01 .42 .08 <.01 <.01 2.55 .03 .06 <.01 .62	153 1050 1530 1350 129 960 960 990 81 1440 1800 1200	<2 <2 <2 <2 <2 <50 <2 <2 <2 <2 <2 <2 <2 <2 <2 <2 <2 <2 <2	DNGRADIENT DNGRADI ENT
S-10A S-10A S-10A S-10A	811119 820317 820622 820922	12.8 11 21 14.4	7.29 7.17 7.30 6.32	5490 4400 9000 3850	.03 .12 <.01 .60	120 570 690 237	8 <2 23 <2	UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT
S-7 S-7 S-7 S-8 S-8 S-8 S-9 S-9 S-9A S-9A S-9A	811119 820317 820622 820922 811119 820317 820622 820922 811119 820317 820622 820922	14.8 10 16 18.8 14.7 9 20 18.3 15.2 11 16 17.8	7.15 6.85 6.89 7.12 7.01 6.83 6.94 7.27 7.02 6.93 6.93	6400 3150 2300 4650 3200 2800 2500 2650 2350 1950 2000 2150	<.01 <.01 <.01241216 <.01221016 <.0135	171 300 105 225 275 99 630 48 99 33 450 75	220 <2 <2 <2 170 130 <2 <2 49 <2 <2 <2	DMGRADIENT
S-12 S-12 S-12 S-12	811112 820315 820622 820922	15 9 11 12.2	6.79 6.78 6.83 6.60	42000 37750 18800 13775	.08 .21 .08 .27	81 420 3000 720	<2 <2 <2 <2	UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT
S-13 S-13 S-13 S-13	811112 820315 820622 820923	13.6 10 13 10.0	7.14 6.91 6.91 6.79	28000 30000 16600 9500	.02 <.01 <.01 .26	186 165 1680 600	2300 4900 8 <2	UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT
S-9 S-9 S-9 S-10 S-10 S-10 S-11 S-11 S-11 S-11	811112 820315 820622 820923 811112 820315 820622 820923 811112 820315 820622 820923	14.4 7 13 11.7 16.1 9 15 14.4 14.3 6 15 12.8	6.95 6.91 6.90 6.73 6.98 6.7 6.78 6.64 6.62 6.62 6.65	18000 18400 10800 13850 39000 40500 20050 22500 65000 61500 68000 49250	.08 .73 <.01 <.01 .06 <.01 <.01 <.01 .14 .18 .07	294 430 60 960 120 145 360 1230 3 385 540 1320	33 <2 <2 <7	DNGRADIENT
S-15 S-15	830315 830628	11 14.4	6.57	8500	<.02 ND	80 0 7 90	<2 ND	UPGRADIENT UPGRADIENT
S-14 S-14	830315 830628	11.5 13.9	6.88	5500	.04 ND	930 710	<2 ND	DNGRADIENT DNGRADIENT
S-16 S-16	830315 830628	14.4			<.01 ND	72 85	33 '	UPGRADIENT UPGRADIENT
S-17 S-17 S-18 S-18	830315 830628 830315 830628	14.4 16.7		-	<.01 ND <.01 ND	1420 1660 1920 1500	5 ND 13 ND	DNGRADIENT DNGRADIENT DNGRADIENT DNGRADIENT

Well	Date	Ag	As	Ba	Cđ	Cr	Fe	Hg	Mn	Pb	Se	Na	Cl	F	Relative
No. S-la	(yymmdd) 811119	(mg/L)	(mg/L)	.060	(mg/L)	(mg/L)	(mg/L)	(mg/L) <0.00020	(mg/L) 1.51	(mg/L)	(mg/L) <.001	(mg/L) 415	(mg/L) 668	(mg/L)	Location UPGRADIENT
S-la S-la S-la	820316 820622 820921	.020 .020 .020	.019 .034 .032	.040 .040 .040	¥.001 .017 .010	.040 <.001 .040	0.090 0.16	<0.00020	2.10 1.450 0.850	.020 .13 .080	<.001 <.001 <.001	425 560 540	381 1305 988	3.40 3.06 2.98	UPGRADIENT UPGRADIENT UPGRADIENT
S-2 S-2 S-2	811119 820316 820622	.020 .012 .020	.176 .129 .146	.075 .060 .060	<.001 <.001 .008	.050 .040 <.001 .010	0.18	<0.00020 <0.00020	0.18 0.090 0.100	.090 .080 .100	<.001 <.001 <.001	635 580 565	1323 2290 2890	4.46 4.82 5.32	DNGRADIENT DNGRADIENT DNGRADIENT
S-2 S-3 S-3	820921 811119 820316	.019 .021 <.001	.110 .084 .007	.090 .110 .100	<.001 <.001 <.001	.030 .030	5.13 0.080	<0.00020 <0.00020 <0.00020	0.060 0.51 0.180	.015 .080 .040	<.001 <.001 <.001	550 420 495	4820 1406 3795	5.16 2.41 2.91	DNGRADIENT DNGRADIENT DNGRADIENT
S-3 S-3 S-4	820622 820921 811119	.002 .002 .020	.085 .074 .069	.060 .110 .050	.005 .001 <.001	<.001 .005 .050		<0.00020 <0.00020 <0.00020	0.265 0.225 1.26	.040 .020 .060	<.001 <.001 <.001	425 570 465	1210 2060 898	3.39 3.34 5.30	DNGRADIENT DNGRADIENT DNGRADIENT
S-4 S-4 S-4	820316 820622 820921	.020 .010 .020	.166 .070 .041	.020 .030 .030	<.001 .006 .001	.040 <.001 .006	0.04 0.11 0.100	<0.00020 <0.00020 <0.00020	0.820 0.890 1.020	.050 .050 .026	<.001 <.001 <.001	455 460 500	1739 1040 2320	5.49 5.35 5.43	DNGRADIENT DNGRADIENT DNGRADIENT
S-10A S-10A S-10A	811119 820317 820622	<.001 <.001 <.001	.018 .005	.085 .220 .180	<.001 <.001 .011	.020 .030 <.001	0.092	<0.00020 <0.00020 <0.00020	0.20 0.140 0.115	.070 .045 .040	<.001 <.001 <.001	865 625 1030	1154.4 437 596.00	3.68 3.19 3.41	UPGRADIENT UPGRADIENT UPGRADIENT
S-10A S-7	820922 811119	<.001	.009	.190	<.001 <.001	<.001		<0.00020 <0.00020	0.110	.040	<.001 <.001	770 1450	1007	3.56 4.43	UPGRADIENT DNGRADIENT
s-7 s-7 s-7	820317 820622 820922	<.001 <.001 <.001	.030 .042	.500 .380 .554	<.001 .004 <.001	.020 .013 <.001	2.65 0.244	<0.00020 <0.00020 <0.00020	0.57 0.4 70	<.001 .030	.003 <.001	490 320	301 217.20	4.68 4.86	DNGRADIENT DNGRADIENT
S-8 S-8	811119 820317	<.001 <.001	.030 .0 36	.210 .300	<.001 <.001	.030 <.001	4.25 0.59	<0.00020 <0.00020	0.375 0.57 0.40	.020 .050 <001	<.001 <.001 <.001	920 460 490	971 294 261	4.10 5.70 5.23	DNGRAÐIENT DNGRAÐIENT DNGRAÐIENT
S-8 S-8 S-9A	820622 820922 811119	<.001 <.001 <.001	.031 .027 .048	.340 .390 .550	.005 <.001 <.001	<.001 <.001 .060		<0.00020 <0.00020 <0.00020	0.360 0.340 0.52	.040 .019 .070	<.001 <.001 <.001	465 555 330	397.00 425 167	5.10 5.49 4.24	DNGRADIENT DNGRADIENT DNGRADIENT
S-9A S-9A S-9A	820317 820622 820922	<.001 <.001 <.001	.016 .018 .023	1.100 1.060 .995	<.001 .004 <.001	.020 <.001 <.001	0.180	<0.00020 <0.00020 <0.00020	0.220 0.190 0.155	<.001 <.001 .040	<.001 <.001 <.001	220 210 250	166 66.40 184	3.70 3.81 3.84	DNGRADIENT DNGRADIENT DNGRADIENT
													4		
S-12 S-12 S-12	811112 820315 820622	.050 .040 .030	.002 .003	.100 .110 .020	<.001 <.001 .035	.080 .075 <.001	10.65	<0.00020 <0.00020 <0.00020	1.25 1.10 0.792	.170 .150 .160	<.001 .002 <.001	6100 4250 3000	5100 5950 4740.00	2.88 3.22 3.50	UPGRADIENT UPGRADIENT UPGRADIENT
S-12 S-13 S-13	820922 811112 820315	.030 .030	.034 .005 .006	.035 .100	.001 .004	.050 .070 .065	18.50	<0.90020 <0.00020 <0.00020	0.950 0.81 0.74	.080 .150 .013	<.001 .020 .020	3700 3700 3400	2950 3827 3809	3.41 2.90 2.80	UPGRADIENT UPGRADIENT UPGRADIENT
S-13 S-13	820622 820923	.020 .020	<.001 <.001	.030 .040	.025 .001	<.001 .040	0.550	< 0.00020 < 0.00020	0.295 0.285	.070 .080	<.001 <.001	1680 3150	4080.00 7200	3.26 3.03	UPGRADIENT UPGRADIENT
s-9 s-9	811112 820315	.030	.025 .021	.100 .095	<.001 <.001	.080 .500	23.15	<0.00020 <0.00020	1.15	.140	<.001 <.001	2100 2350	2532 3164	3.64 3.15	DNGRADIENT DNGRADIENT
S-9 S-9 S-10	820622 820923 811112	.020 <.001 .070	.055 .012 .014	.060 .020 .150	.022 <.001 <.001		6.650 110.00	<0.00020 <0.00020 <0.00020	0.240 0.230 1.43	.080 .060 .660	<.001 <.001 <.001	60 3150 5100	2690.00 5150.00 4729	3.21 3.20 2.63	DNGRADIENT DNGRADIENT DNGRADIENT
S-10 S-10 S-10	820315 820622 820923	.090 .030 <.001	.048 .106 .015	.130 .030 .025	<.001 .020 <.001	.250 .003 .044		<0.00020 <0.00020 ND	1.30 0.680 0.560	.250 .100 .010	<.001 <.001 <.001	5200 360 5400	6450 7610.00 9710.00	2.67 2.61 2.61	DNGRADIENT DNGRADIENT DNGRADIENT
S-11 S-11 S-11	811112 820315 820622	.090 .090	009. 800. 100.>	.100 .114 .030	<.001 <.001 .080	.012 .120 <.001	6.75	<0.00020 <0.00020 <0.00020	0.95 0.85 0.530	.300 .220 .270	<.001 <.001 <.001	13500 14000 540	6780 8500 17900.00	1.12 1.34 1.25	DNGRADIENT DNGRADIENT DNGRADIENT
S-11	820923	<.001	.017	.010	.005	.100		<0.00020	0.490	.032	<.001	13500	20760.00	1.29	DIGRADIENT
S-15 S-15	830315 830628	<.001 ND	.013 .013	.42 .410	<.001 ND	<.001 .008	0.130 0.140	ND	0.99 0 0.740	.007 .002	<.001	1810 1740	1851 2600	2.94 3.15	upgradient upgradient
S-14 S-14	830315 830628	<.001 ND	.007 .020	.11 .150	.004 .003	.002 .008	0.270 0.180	ND	0.105 0.130	.004 .009	<.001 ND	967 93	1404 1280	3.51 3.64	DNGRADIENT DNGRADIENT
	A4	· ·		. -	<u>-</u>		A :-		- ·-			.=.			
S-16 S-16	830315 830628	.004 ND	<.001 ND	.06 .090	.009 ND	800. CM	0.45 0.590	ND	0,17 0,310	.005 ND	VID .>>	450 438	2062 2030	2.89 2.78	UPGRADIENT UPGRADIENT
S-17 S-17 S-18	830315 830628 830315	.016 .002	<.001 .010 <.001	.05 .030 .05	.010 .002 .012	.026 .011 .020	0.31 0.300 0.288	ND	0.65 0.250 0.35	.066 .002 .070	<.001 ND <.001	3450 605 2160	1569 8500 1174	2.01 1.75 3.83	DNGRADIENT DNGRADIENT DNGRADIENT
S-18	830628	ИĎ	.156	.060	.005	.012	0.160	ND - 53 -	0.200	.005	ND	205	280	4.05	DNGRADIENT
								- 53,							

TABLE 5. (cont'd)

Well No.	Date (yymmdd)	TOC (mg/L)	TOX (mg/L)	Phenol (mg/L)	Gross Alpha (pCi/L)	Gross Beta (pCi/L)	Radium (pCi/L)		Lindane (mg/L)	Methoxy- chlor (mg/L)	Toxa- phene (mg/L)	2,4-D : (mg/L)	2,4,5-TP (mg/L)	Relative Location
S-1A S-1A S-1A S-1A	811119 820316 820622 820921	59.50 76.20 65.70 69.10	20.50 4.20 3.100 2.200	.114 .058 .058 .060	1.5 1.6 <2.0 <2.0	<0.1 22. 16.0 < 0.1	<0.1 <0.1 <0.1 <0.1	<.0001 <.0001 <.0001 <.0001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT
5-2 5-2 5-3 5-3 5-3 5-3 5-4 5-4 5-4	811119 820316 820622 820921 811119 820316 820622 820921 811119 820316 820622 820921	137.40 120.70 116.00 121.50 148.30 115.80 134.70 140.50	26.3 21.500 20.200 24.20 25.50 24.300 25.500 25.60	.017 .026 .068 .588 .016 .020 .015 .138 .013 .017	1.2 1.6 <2.0 <2.0 1.5 1.4 <2.0 <1.0 1.3 <2.0 <2.0	<0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1	<0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1	<.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	DNGRADIENT
S-10A S-10A S-10A S-10A	811119 820317 820622 820922	71.9 68.90	19.80 1.95 2.000 12.00	.140 .173 .145 .425	2.5 1.5 <2.0 <2.0	<0.1 29.0 12.0 <0.1	<0.1 <0.1 <0.1 <0.1	<.0001 <.0001 <.0001 <.0001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001	UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT
S-7 S-7 S-7 S-8 S-8 S-8 S-9 S-9A S-9A S-9A	811119 820317 820622 820922 811119 820317 820622 820922 811119 820317 820622 820922	52.80 57.40 26.50 77.80 89.60 97.40 25.10 39.00 38.40	13.10 12.90 8.300 7.10	.103 .114 .093 .233 .030 2.79 1:20 1.750 .033 .045 .163	1.9 <2.0 <2.0 <2.0 1.6 3.8 4.0 <2.0 1.4 1.4 <2.0 <2.0	<0.1 2.0 <2.0 <0.1 <0.1 2.2 <2.0 <0.1 <0.1 2.7 <2.0 <0.1	<0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1	<.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	DNGRADI ENT DNGRADI ENT
S-12 S-12 S-12 S-12 S-13 S-13 S-13	811112 820315 820622 820922 811112 820315 820622 820923	40.90 55.40 57.10 44.00 31.50 171.40 245.60 32.00	4.50 6.40 0.560 0.550 3.50 3.10 0.680 0.540	<.001 .009 .050 .033 <.001 <.001 .042	1.7 1.8 <2.0 <2.0 1.6 1.4 <2.0 <2.0	<pre><0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1</pre>	<0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1	<.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001	<.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001	UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT UPGRADIENT
S-9 S-9 S-9 S-10 S-10 S-11 S-11 S-11 S-11	811112 820315 820622 820923 811112 820315 820622 820923 811112 820315 820622 820923	42.00 145.30 201.20 96.80 185.00 64.70 73.00 53.20	13.50 12.600 10.100 9.50 9.10 10.200 7.500 24.50	.014 .018 .030 .058 .006 .013 .030 .129 .004 .010	1.6 27.0 12.0 18. 1.4 27.0 <2.0 20. 1.7 <2.0 <2.0 <2.0	3.6 1.2 < 2.0 1.5 2.8 1.5 < 2.0 1.3 3.4 1.8 < 2.0 < 0.1	<0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1 <0.1	<.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001 <.0001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	<.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001 <.001	DNGRADIENT
S-15 S-15	830315 830628			.204 .160	< 2.0 ND	<0.1 ND	<0.1 ND	<.0001 ND	<.001 ND	<.001 ND	<.001 ND	<.001 ND	<.001 ND	UPGRADIENT UPGRADIENT
S-14 S-14	830315 830628			.167 .150	< 2.0 ND	<0.1 ND	<0.1 ND	<.0001 ND	<.001 ND	<.001 ND	<.001 ND	<.001 ND	<.001 ND	DNGRADIENT DNGRADIENT
s-16 s-16	830315 830628			.006 .004	< 2.0 ND	< 0.1	<0.1 ND	<,0001 ND	<.001 ND	<.001 ND	<.001 ND	<.001 ND	<.001 ND	UPGRADIENT UPGRADIENT
S-17 S-17 S-18 S-18	830315 830628 830315 830628			.050 .046 .058 .036	< 2.0 ND < 2.0 ND	<0.1 ND <0.1 ND	<0.1 ND <0.1 ND	<.0001 ND <.0001 ND	<.001 ND <.001 ND	<.001 ND <.001 ND	<.001 ND <.001 ND	<.001 ND <.001 ND	<.001 ND <.001 ND	DNGRADIENT DNGRADIENT DNGRADIENT DNGRADIENT

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constituents present except for TOC. Other parameter values do not indicate a substantial increase or decrease in levels.

Quality data near the oxidation and drying beds "upgradient" wells S-10A and S-15 and "downgradient" wells S-7, S-8, S-9A and S-14 indicate that well S-10A could very well be a downgradient well. If well S-10A is indeed an upgradient well, the data indicate a substantial decrease in the levels of the same parameters, with the exception of C1, which shows a significant increase in levels near the stormwater impoundment area. Other parameter values do not indicate a substantial increase or decrease in levels. If well S-10A is indeed a downgradient well, there is not enough data to make a first-cut analysis.

2. Vitro Uranium Mill Tailing Site

Shallow groundwater quality data at the Vitro Site, as covered by this report, was collected during 1983 from 23 monitoring wells. Samples were collected by the City-County Health Department and analyzed at the State Health Department laboratory. Wells were installed by private contractors conducting studies for the Department of Energy. The Vitro Site is an abandoned uranium mill site and the remains (mill tailings with varying degrees of activity) are piled about the area. Deep wells (115-137 feet) were stationed around the perimeter of the site. Shallow wells (10-41.5 feet) were also stationed around the perimeter in addition to being clustered for sampling aquifers at specific depths. Physical well data is listed in Tables 6 and 7. Locations of wells are shown in Figure 19. Groundwater depth data is not available at this time.

Implications based on 1983 City-County/State Health data are discussed below. The small number of samples, especially at upgradient shallow wells, leads to a high variability in the means of constituent values. As with other

TABLE 6. Physical Well Data; Vitro Uranium Mill Tailings Site - Jacobs Wells

VITRO WELL DATA - JACOBS

				_	
Well No.	Total Depth (ft.)	Screened Interval (ft.)	Blank Interval (ft.)	Ground Elevation (ft.)	Location*
1	115	100-110	0-100	4244.96	ט
2	135	119-129	0-119	4238.07	Ū
3	125	110-120	0-110	4234.13	С
4	137	105-115	0-105	4234.37	D
5	132	115.5-125.5	0-115.5	4234.92	D
6A 6B 6C	31.5 20 10	26-28 17-19 7-9	0-26 0-17 0-7	4244.61 4244.68 4244.71	ប ប ប
7A 7B 7C	35 25 15	26-28 15-17 8-10	0-26 0-15 0-8	4237.97 4238.02 4237.98	ט ט
8A 8B 8C	40 35 10	26-28 19-21 6-8	0-26 0-19 0-6	4233.13 4233.13 4233.13	ccc
9A 9B 9C	35 25 15	27-29 16-18 6-8	0-27 0-16 0-6	4233.17 4232.97 4233.02	D D
10A 10B 10C	36.5 20 10	26-28 14.5-16.5 6-8	0-26 0-14.5 0-6	4234.60 4234.65 4234.55	D D D
11A 11B 11C	31.5 20 15	24-26 17-19 6-8	0-24 0-17 0-6	4234.52 4234.62 4234.82	D D D
12A 12B 12C	31.5 20 10	26-28 17-19 6.5-8.5	0-26 0-17 0-6.5	4235.90 4236.06 4235.95	ccc
13A 13B 13C	31.5 20 10	26-28 16-18 6-8	0-26 0-16 0-6	4235.70 4235.75 4235.75	ם ס ס
14	41.5	18-38	0-18	4233.27	D
15	133	102-112	0-102	4233.73	D

Note: All depths from ground surface.

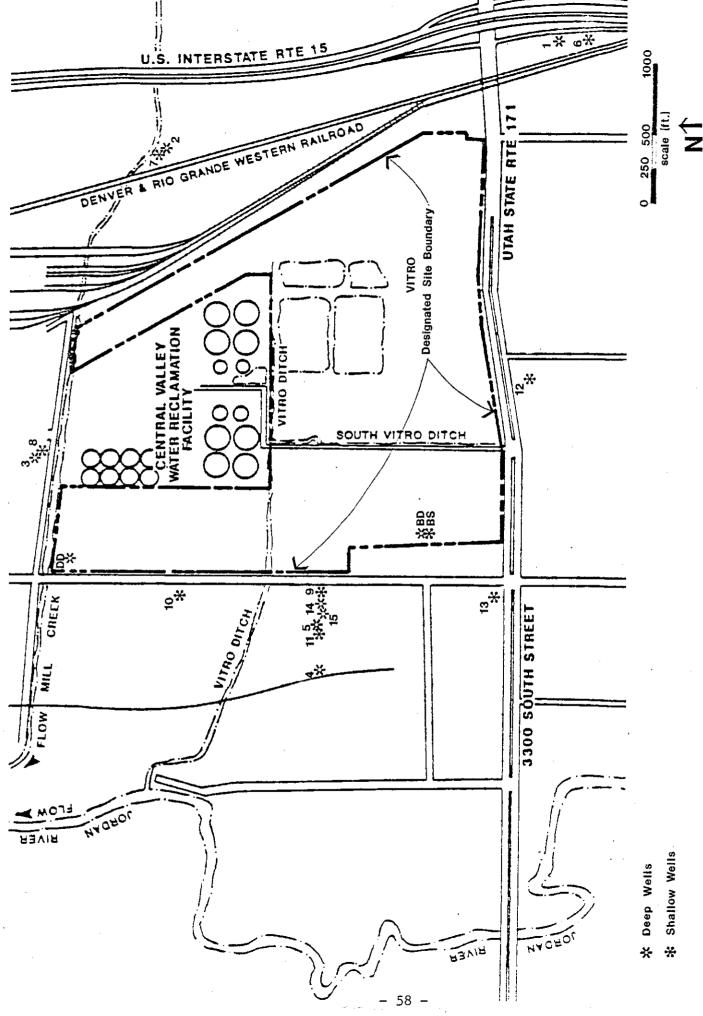
^{*} U - upgradient C - crossgradient D - downgradient

TABLE 7. Physical Well Data; Vitro Uranium Mill Tailings Site - Dames & Moore Wells

VITRO WELL DATA - DAMES & MOORE

Well No.	Blank Interval (ft.)	Screened Interval (ft.)	Sand Pack Interval (ft.)	Grout Seal Interval (ft.)	Bentonite Seal Interval (ft.)
V-BD	0.0-85.0 95.0-98.0	85.0-95.0	82.0-97.0	0.0-80.0	80.0-82.0
V-BS	0.0-15.0 20.0-25.0	15.0-20.0	14.5-25.0	0.0-13.5	13.5-14.5
V-DD	0.0-80.0	80.0-90.0	75.0-90.0	0.0-73.0	73.0-75.0

Note: All depths from ground surface.



Location of Monitoring Wells; Vitro Uranium Mill Tailings Site FIGURE 19.

studies presented in this report, more data is necessary to more accurately describe the long-range implications in groundwater quality.

Groundwater quality data discussed in this report is listed in Table 8. There is no substantial decrease or increase in parameter concentration as groundwater traverses the site for all constituents collected from shallow wells except for As, Cr, Fe, Mn, U, gross alpha and gross beta. Apparently concentrations of As, actually decrease by about 1/3. However, Cr and Mn concentrations approximately double, Fe concentrations increase by about 20%, and gross alpha and beta concentrations increase by an order of magnitude (ten times). U concentrations also increase (somewhat dramatically) but a very small number of upgradient samples could skew the analysis.

As stated above, much more sampling needs to be undertaken to better define the long-term impacts of the vitro remains.

Salt Lake City-County Landfill

Shallow groundwater quality monitoring near the current (post-1982) and old pre-(1982) Salt Lake City-County landfills began in late 1982. To date, two sampling runs have been made with a third scheduled for November 1983. Samples are collected from shallow monitoring wells (less than 25 feet) and are analyzed by the Salt Lake City-County Health Department. The wells are located up and downgradient from both sites. Well locations are shown in Figure 20.

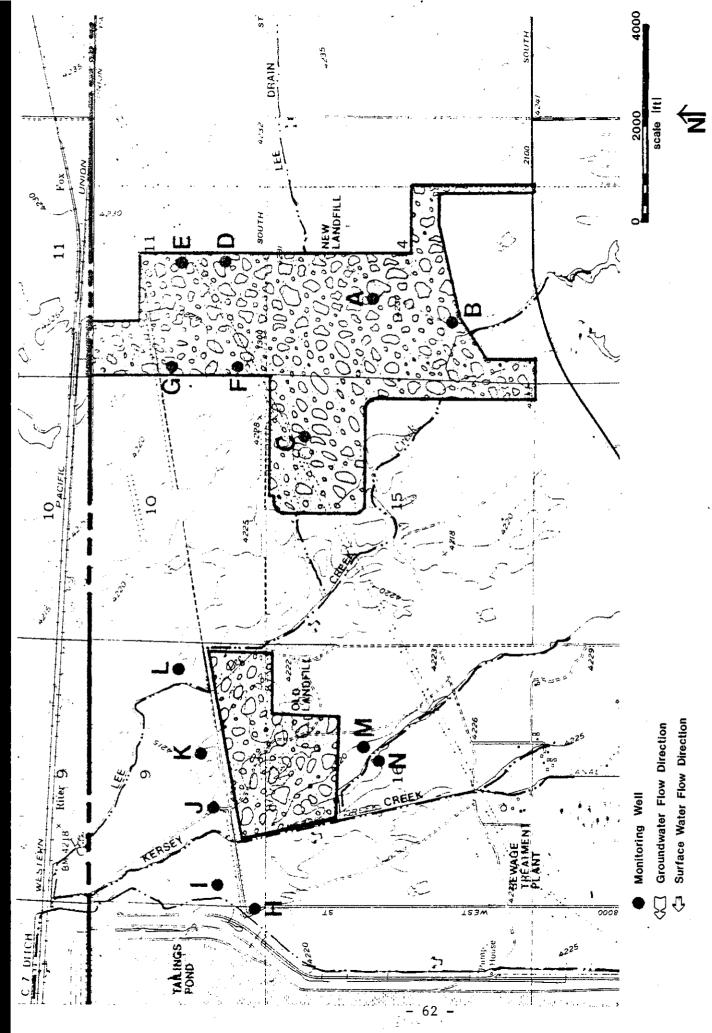
Monitoring wells consist of 4-inch schedule 40 PVC pipe installed to a depth of 10 feet below water table. The deepest well is approximately 25 feet. The 10 feet extending below water table is slotted at 6 inch intervals

TABLE 8. Shallow Groundwater Quality - Vitro Uranium Mill Tailings Site

Well Wo.	Date (yymmdd)	Time	Depth	As (ug/L)	Ba (mg/L)	Cđ (ug/L)	Cr (ug/L)	Cu (ug/L)	Fe (mg/L)	Pb (ug/L)	Min (ug/L)
:	830312	1400	Deep	33	.14	<1	5	< 10	.23	<5	105
2	830512	1200	Deep	22	.11	< 1	< 5	<10	.66	< 5	110
3	830511	1330	Deep	1.0	.10	< 1	<5	<10	24	< 5	65
4C 4	830609 830822	1230 1315	10' Deep?	12 14	<.05 .07	<1 <1	15 < 5	<10 <10	.40 2.20	13 < 5	815 225
5 5	83050 9 830822	1115	Deeb	1.5 1.5	.22 .15	<1<1	< 5 < 5	<10 15	.54 · .30	< 5 < 5	335 310
6A 6A	830409 830413	1300 1400	30 '	14 133	<.05 .28	< 1 < 1	5 < 5	<10 <10	8.90 <.20	< 5 < 5	1010 1 6 0
7 7C 7A	830330 830608 830609	1730 1345 0920	Shal 10' 30'	70.0 162 245.0	.05 .06 .07	< 1 < 1	< 5 6 6	<10 <10 <10	5.00 3.40 .03	< 5 7 < 5	600 415 60
8B	830311	1130	20 •	3.0	<.05	< 1	< 5	<10	14.00	< 5	515
9B 9C	830408 830408	1100 0945	20° 10°	1 3	<.05 <.05	<1 <1	<5 <5	<10 <10	<.03 .49	< 5 < 5	3145 1180
10 10 10 10A	830329 830607 830607 830608	1245 1400 1450 1150	10 ' 10 ' 20 ' 30 '	11.0 21.0 14.0 4	<.05 <.05 <.05 <.05	1 2 2 4	6 14 11 13	15 10 <10 <10	.36 <.03 18.50 2.65	< 5 8 5 6	360 590 2590 865
11A 11D	830609 830609	1120 1030	30' Shal	13.0 35	<.05 <.05	<1 <1	30 30	< 10 < 10	18.40 21.40	6 < 5	1285 1655
120	830517	1145	10'	20	<.05	<1	< 5 -	< 10	<.03	< 5	700
13 13	830406 830517	0830 1345	Shal Shal	525 <.5	.07 .40	<1 <1	5 <5	< 10 10	<.03 .41	< 5 < 5	<10 315
14 14	830602 830602	0010 1320	Shal Shal	3.0 3	<.05 <.05	5 4	<5 <5	<10 <10	14.80 <.03	<5 <5	1670 1740
15 15 15	830521 830523 830823	1715 0920 1025	Deep Deep	<.5 <.5 <.5	.30 .37 .31	<1 <1 <1	6 < 5 < 5	< 10 < 10 < 10	1.50 .08 1.80	<5 <5 <5	415 420 380
BD	83060 6	1320	951	13.0	.34	<1	5	< 10	.60	<5	115
DD	830606	1205	901	<.5	.17	<1	<5	< 10	<.03	<5	25
DD	830607	1150	35'	190.0	<.05	<1	<\$	< 10	8.67	<5	745
BS	830610	0945	?	167.0	<.05	3	< 5	< 10	13.00	7	1780

TABLE 8. (cont'd)

			•								•	
Well No.	Date (yymmdd)	Time	Hg (ug/L)	Ni (ug/L)	Se (ug/L)	Ag (ug/L)	Zn (ug/L)	Mo (ug/L)	Gross Alpha (pCi/L)	Gross Beta (pCi/L)	Radium- 226 (pCi/L)	U (ug/L)
1	830312	1400	<0.1	< 10	<.5	<2	<5		4	<10.0	<0.5	0
2	830512	1200	<0.1	< 10	<.5	<2	<5		<2.0	11	<0.5	1
3	830511	1330	<0.1	<10	<.5	<2	<5		<2.0	<10.0	<0.5	0
4C 4	830609 830822	1230 1315	<0.1 <0.1	185 <10	<.5 <.5	< 2	10 10	< 50	303 2	763 <10.0	1	693 1
5 5	830509 830822	1115	<0.1 <0.1	<10 <10	<.5 <.5	<2 <2	<5 <5	< 50	<2.0	<10.0	<0.5	0
6A 6A	830409 830413	1300 1400	<0.1 <0.1	86 <10	<.5 <.5	<2 <2	40 < 5					
7 7C 7A	830330 830608 830609	1730 1345 0920	<0.1 <0.1 <0.1	53 <10 <10	<,5 <,5 <,5	<2 <2 <2	5 < 5 < 5		25 <2.0	298 91	<0.5 <0.5	4 2
8B	830311	1130	<0.1	19	<.5	<2	< 5		5	<10.0	0	9
9B 9C	830408 830408	1100 0945	<0.1 <0.1	7 4 7 7	<.5 <.5	3 <2	20 50		63 818	<10.0 513	1 <0.5	59 1350
10 10 10 10A	830329 830607 830607 830608	1245 1400 1450 1150	<0.1 <0.1 <0.1 <0.1	44 86 53 <10	<.5 .5 <.5 <.5	<2 4 <2 6	< 5 5 < 5 < 5		59 42 21 19	156 56 420	<0.5 <0.5 <0.5	84 68 15 2
11A 11D	830609 830609	1120 1030	<0.1 <0.1	430 385	<.5 <.5	6 6	15 15		63 142	770 1093	<0.5 <0.5	78 110
12C	830517	1145	0.1	<10	<.5	₹2	20	•	6	<10.0	<0.5	16
13 13	830406 830517	0830 1345	<0.1 0.1	<10 <10	<.5 <.5	<2 <2	< 5 < 5		8 9	8 9 12	<0.5 1	13 0
14- 14	830602 830602	0010 1320	<0.1 <0.1	<10 <10	<.5 <.5	4	<5 110		124 362	42 127	<0.5 <0.5	112 78
15 15 1 5	830521 830523 830823	1715 0920 1025	<0.1 <0.1 <0.1	<10 <10 <10	<.5 <.5 <.5	<2 <2 <2	<5 <5 <5	<200 <200 <50	7 7 <2.0	36 11 <10.0	1 <0.5 <0.5	1 0 0
BD	830606	1320	<0.1	<10	< .5	<2	<\$		3	67	<0.5	0
DD	83060 6	1205	<0.1	<10	<.5	<2	<5		<2.0	49	<0.5	0
ממ	830607	1150	<0.1	27	<.5	<2	5		130	<10.0	<0.5	461
BS	830610	0945	<0.1	< 10	<.5	<2	20		139	32	0.5	20 2



Location of Monitoring Wells; Salt Lake City - County Landfill FIGURE 20.

staggered on two sides, wrapped with fiber screen cloth and backfilled with pea gravel (3/8 inch maximum). The bottom is capped with a bonded PVC cap. The upper portion of the casing (ground elevation to water table) is not slotted and is backfilled with native soil compacted to 90% (ASTM). A one-foot minimum depth concrete seal caps the bore hole. A steel cover with a locked hasp secures the well. Depth of individual wells and depth to groundwater are not available at this time. Groundwater depths will be measured during the November sampling run.

Water quality data for the two sampling runs is shown in Table 9. A cursory analysis of data was made by the Bureau of Water Quality, City-County Health Department for the County Public Works Department (operators of the landfills). A copy of this analysis is included in the appendix. Based on one or two samples per well, it is difficult to make sound quality judgments. Therefore, no further analysis of data is made in this report.

TABLE 9. Shallow Groundwater Quality - Salt Lake City-County Landfill

Well No.	Date (yymmdd)	Temp (C)	TDS (mg/L)	As (mg/L)	Cđ (mg/L)	Cr (mg/L)	Fe (mg/L)	Hg (mg/L)	Mg (mg/L)	Pb (mg/L)	Se (mg/L)	Zn (mg/L)	Na (mg/L)
A A	821110 830706	11.2 17.2	9,500	0.38	<0.02 <0.002	<0.02 <0.002	0.24 0.08	0.001	40 10	<0.01 0.005	<0.5	<0.05 <0.02	3,750 3,000
B B	821110 830706	11.4 13.6	20,000	0.02	<0.02 <0.002	<0.02 0.002	0.50 3.10	<0.001 <0.001	42 38	0.02 0.013	< 0.5	<0.05 <0.02	7,750 1,000
c c	821110 830706	12.2 15.8	3,900	0.38	<0.02 <0.002	<0.02 0.002	0.23 0.39	<0.001 <0.001	78 19	0.01 0.008	<0.5	<0.05 0.03	1,200 8,000
D D	821110 830 706	12.6 17.2	13,000	0.01	0.04 <0.002	<0.02 <0.002	2.20 0.79	<0.001 <0.001	175 35	0.04 0.005	<0.5	0.49 <0.02	4,250 2,150
E E	821110 830706	12.4 14.0	12,000	0.05	0.03 0.002	<0.02 <0.002	0.18 1.65	<0.001 <0.001	130 75	<0.01 0.027	1.0	<0.05 0.02	3,750 4,200
P P	821110 830706	11.8 15.2	22,000		<0.02 0.002	<0.02 0.003	0.16 7.50	<0.001	26 88	<0.01 0.009	<0.5	0.05 0.16	9,000 3,200
G G	821110 830706	14.5 16.8	12,500	0.34	<0.02 <0.002	<0.02 0.002	2.50 0.01	<0.001 <0.001	100 34	0.03 0.008	<0.5	0.29 0.02	4,250 2,300
H	821110 830706	12.5 14.6	7,000	0.24	<0.02 <0.002	<0.02 <0.002	0.16 0.60	<0.001 <0.001	105 40	<0.01 0.003	<0.5	<0.05 <0.02	2,500 2,300
I I	821110 830706	11.8 14.2	7,500	0.28	<0.02 0.004	<0.02 <0.002	4.00 6.50	<0.001 <0.001	115 18	<0.01 0.005	<0.5	<0.05 0.05	2,600 1,950
J J	821110 830706	10.8 14.2	22,000	0.18	<0.02 <0.002	<0.02 <0.002	0.16 0.04	<0.001 <0.001	25 10	<0.01 0.003	1.0	0.05 <0.02	8,750 9,400
ĸ	830706	15.4		0.13	0.002	0.002	0,53	<0.001	14	0.004	<0.5	<0.02	970
L	830706	15.0		0.03	<0.002	<0.002	0.28	<0.001	0.7	0.002	0.5	<0.02	1,500
M M	821110 830706	12.5 13.0	18,000	0.21	<0.02 <0.002	<0.02 <0.002	0.12 0.68	<0.001 <0.001	110 25	<0.01 0.011	1.0	<0.05 0.04	7,250 3,500
N	830706	13.2		0.23	<0.002	<0.002	0.74	<0.001	33	0.007	<0.5	<0.02	5,000

TABLE 9. (cont'd)

Well No.	Date (yymmdd)	Cl (mg/L)	NO2-N (mg/L)	NO3-N (mg/L)	PO4-P (mg/L)	SO ₄ (mg/L)	TOC (mg/L)	DO (mg/L)	O&G (mg/L)	F. Coli (MPN/100ml)	T. Coli (MPN/100ml)
A A	821110 830706	4,400 2,950	<0.02 0.044	0.83 1.93	0.4 0.63	1,500 1,000	28.9	2.6 4.4	0.6	· — <2	20
B	821110 830706	9,000 1,250	<0.02 0.580	1.70 0.19	0.9 0.13	3,300 240	3.6	3.4 2.2	<0.1	<2	<10
C C	821110 830706	1,800 8,600	<0.02 0.014	0.53 1.60	<0.01 0.31	300 2,800	27.1	3.0 4.6	0.3	-	<10
D D	821110 83070 6	7,500 3,600	<0.02 0.154	3.00 0.00	<0.1 0.11	575 300	2.7	8.2 3.6	0.3	<2	80
E E	821110 830706	6,500 6,350	<0.02 0.400	1.00 0.96	<0.1 0.37	600 1,100	6.8	2.2 3.3	0.2	<2	400E
P P	821110 830706	11,000 6,550	<0.02 0.790	2.40 0.00	0.8 0.03	2,300 250	1.0	3.6 4.2	1.2	- <2	20
G G	821110 830706	7,000 3,050	< 0.02 0.580	0.28 0.50	0.1 0.20	350 1,100	176.8	6.6 2.9	1.1	<2	UNS.
H	821110 830706	2,750 2,250	<0.02 1.180	1.20 0.93	0.4 0.59	1,400 1,700	9.0	3.4 3.0	0.9	<2	< 2
I	821110 830706	2,900 2,300	<0.02 0.053	1.10	<0.1 0.42	1,200 600	79.9	2.5 1.8	0.5	<2	<2
J J	821110 830706	11,000 9,700	<0.02 0.010	2.20 1.41	0.7 0.78	2,300 1,300	149.9	2.5 2.0	4.6	<2	< 2
ĸ	830706	1,200	0.044	0.86	0.39 -	120	32.1	1.6	·	<2	<2
L	830706	1,950	0.018	0.55	0.09	850	14.1	2.2	-	16	<2
M M	821110 83 0706	9,000 3,400	<0.02 1.240	0.76 0.12	0.1 0.48	1,900 300	4.8	2.8 1.6	1.9	32	- <2
n	830706	6,750	1.210	0.31	0.47	1,000	13.3	1.8	—	<2	16

REFERENCES

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- 3. State of Utah, Office of the Governor, Salt Lake County 208 Plan Certification, April 22, 1979
- 4. David W. Miller, "Groundwater Contamination Fundamentals and Monitoring." U.S. EPA Office of Solid Waste. 1978
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- 6. Steven F. Jensen and Peter S. Borromeo, Revised Workplan for Groundwater Quality/Quantity Assessment for Salt Lake County, Division of Water Quality and Pollution Control. July, 1981
- 7. Utah State Department of Social Services, Division of Environmental Health, Standards of Quality for Waters of the State, Part II, Waste Disposal Code. Amended, 1978
- 8. Paula L. Magnuson, "Groundwater Classification" Region VIII EPA Groundwater Seminar, July, 1981
- 9. Kidd Waddell, U.S. Geological Survey. "Personal Communication" 1982
- 10. U.S. Environmental Protection Agency, <u>Interim Guidelines and Specifications for Preparing Quality Assurance Project Plans</u>, Office of Research and Development, Washington, D.C. December, 1980
- 11. U.S. Environmental Protection Agency, Region VIII, The Quality Assurance Program Plan for EPA Region VIII Denver, Colorado. January, 1982
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- 13. Utah State Health Laboratory, "Quality Assurance Program Plan" Bureau of Environmental Chemistry, July, 1983
- 14. United States Dept. of Agriculture Soil Conservation Service, Soil Survey of Salt Lake Area, Utah. April, 1974
- 15. Utah Geological and Mineralogical Survey, Depth to Shallow Groundwater, by Ben Everett. 1982
- 16. State of Utah Dept. of Natural Resources, Water Resources of Salt Lake County, Utah, USGS Technical Publication 31. 1971
- 17. Ibid.
- 18. R.L. Seiler and K.M. Waddell, <u>Reconnaissance of Shallow-Unconfined Aquifer in Salt Lake Valley</u>, Utah, U.S. Geological Survey, Salt Lake City, Utah (Unpublished report).
- 19. U.S. Environmental Protection Agency, Region VIII "Uncontrolled Hazardous Waste Disposal Site; Rose Park, Salt Lake County, Utah", Denver CO, 1983.

APPENDIX A

APPENDIX A IS AVAILABLE IN FINAL PRINTED FORM FROM THE U.S. GOVERNMENT PRINTING OFFICE OR THE U.S. GEOLOGICAL SURVEY

THE DOCUMENT CAN BE OBTAINED UNDER THE TITLE OF RECONNAISSANCE OF THE SHALLOW-UNCONFINED AQUIFER IN SALT LAKE VALLEY, UTAH*

WATER RESOURCES INVESTIGATIONS REPORT #

83-4272

For additional information write to:
District Chief, U.S. Geological Survey, WRD
Room 1016 Administration Building
1745 West 1700 South
Salt Lake City, Utah 84104

Copies of this report can be purchased from:
Open-File Services Section, Western Distribution Branch
U.S. Geological Survey
Box 25425, Federal Center, Lakewood, Colorado, 80225
(Telephone 303-234-5888)

Appendix B



610 South 2nd East, Salt Lake City, Utah 84111 Phone: 532-2002

> HARRY L. GIBBONS, M.D., M.P.H. Director

11 October, 1983

Romney Stewart Associate Public Works Director 2033 South State Street Salt Lake City, Utah 84115

Dear Romney:

The complete samples results for July 6, 1983 of the Salt Lake County Landfill Groundwater Monitoring Wells have now been received. Some of the results have changed since the samples of November 10, 1982. These will be discussed in this letter.

The bacteriological results are not comparable to the November 10, 1982 samples, since the 1982 samples were contaminated, and some results were overgrown. The bacteriological samples of July 6, 1983 will have to be used as a background for all future samples.

It is anticipated that during November, 1983 all of the Kennecott area wells will again be sampled.

Chemical analyses are compared as follows:

Well C showed an increase in Chloride from 1800~mg/l to 8600~mg/l. All other wells were very similar, or dropped in Chloride level since the 1982~samples.

Nitrite levels (NO as N) increased over 1982 mainly in wells E, F, G, H, KSL-1, KSL- 2 and KSL-3. The increase in KSL-3 is not significant since KSL-1 and KSL-2 are upgrade from KSL-3 and they increased at approximately the same ratio.

Well C had a great increase in sulfate (SO $_4$) from 300 mg/l to 2800 mg/l. An increase in Sodium from 1200 mg/l to 8000 mg/l was also noted in Well C.

Well F showed the greatest increase in Magnesium from 26 mg/l to 88 mg/l. All other wells either decreased, or were close to the 1982 results.

BOARD OF HEALTH
Wayne I. Baer, Chairman
L. Jed Morrison, M.D.
Vice Chairman
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City Government
M. Tom Shimizu
County Commissioner
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John M. Bevan, D.D.S.

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LaRell D. Muir, Mayor

W. Paul Thompson Sandra K. Ercanbrack Secretary

Murray City Rulon Simmons, M.D. L.C. Romney

Jeri Taylor

EIV CC Condidi

Iron increased in Well B (.50 mg/l to 3.10 mg/l) and in Well F (.16 mg/l to 7.50 mg/l).

Zinc, Cadmium, Lead, Chromium and Mercury remained relatively the same in both samplings.

The majority of the results of the parameters tested decreased from 1982 to 1983. Perhaps the different levels of groundwater from 1982 to 1983 may have had some influence in the decrease in some parameter results, whereas more dissolving of Sodium, Sulfate, Chloride and Nitrate could have taken place. A better pattern may be seen after the next sampling results are compiled.

After reviewing your copy of the results, if you have any questions, please contact our office at 532-2002 Ext: 543.

Sincerely,

SALT LAKE CITY-COUNTY HEALTH DEPARTMENT

Frank V. Nabrotzky, Environmental Health

Quality Specialist

Bureau of Water Quality

FVN/pn

Encl: 3

Appendix C

Scott M. Matheson Governor

STATE OF UTAH DEPARTMENT OF HEALTH

STATE HEALTH LABORATORY

44 Medical Drive, Salt Lake City, Utah 84113

Francis M. Urry, Ph.D., Director Room 207 801-533-6131



James O. Mason, M.D., Dr.P.H. Executive Director 801-533-6111

DIVISIONS

Community Health Services Environmental Health Family Health Services Health Care Financing and Standards

OFFICES

Administrative Services Health Planning and Policy Development Medical Examiner State Health Laboratory September 13, 1983

RECEIVED

Peter Borromeo

Salt Lake County Flood Control

2033 South State Street

Salt Lake City, Utah 84115

SEP 1 6 1983

SALT LAKE COUNTY FLOOD CONTROL & WATER QUALITY

Dear Mr. Borromeo:

On or about August 24, 1983 we reported results for our sample number 832781, your field number 1019. Since that time, we have found contamination in our filtration system affecting mineral analysis.

The sample in the original container was reanalyzed with the following results:

Specific Conductivity 4.5µmhos/cm Total Dissolved Solids <5mg/1Chloride <1PPm Sulfate <10PPM Arsenic <0.5PPb Cadmium <1.0PPb Chromium <5.0PPb Lead <5.0PPb

We are sorry for any inconvenience the previous data may have caused your office.

Sincerely,

Ralph A. Helfer

Chemist

Quality Assurance Section

Bureau of Laboratory Improvement



UNITED STATES ENVIRONMENTAL PROTECTION AGENCY

REGION VIII

1860 LINCOLN STREET

DENVER, COLORADO 80295

MAY 22 1984

neo-entro

MAY 24 1984

SALT LAKE COUNTY
FLOOD CONTROL & WATER GUALITY

Mr. Steven F. Jensen Water Quality Program Manager Salt Lake County Flood Control and Water Quality Division 2033 South State Street Salt Lake City, Utah 84115

> RE: Draft Assessment of Shallow Aquifer Quality in Salt Lake County

Dear Steve:

Ref: 8WM-SP

EPA has reviewed the referenced Section 208 assessment. It represents a commendable inter-agency effort which we hope will continue to be strengthened. Our comments are attached.

The assessment should prove to be valuable not only as a basis for longer term monitoring, but also as a basis for current ground water protection actions. If you have any questions, feel free to call.

Sincerely yours,

Doug Lofstedt

Utah Areawide Project Officer

Attachment

cc: Roy Gunnell

Utah Bureau of Water Pollution Control

ATTACHMENT

EPA Comments on Draft Assessment of Shallow Aquifer Quality in Salt Lake County (October 1983)

Under Section III on page 5, we suggest that the last sentence of the fifth paragraph be clarified in a couple of areas. Many readers may not be familiar with the "solute transport" process. It also isn't clear what is being referred to by the "critical limits to this deep/shallow aquifer interface" which will be prescribed by computer modelling.

Section III should also include a brief rationale for selection of parameters and compounds which were analyzed.

We suggest the inclusion of more detailed soils information, particularly the organic content and local modifiers of pH. This would be needed to predict the mobility of hazardous compounds in the soil and water.

The symbols used on page 29 for identifying the wells should be described in the legend.

The assessment should be more specific on NPS identification, continued assessment, and implementation action now needed. What are the "certain locations" where nonpoint contaminant sources may be impairing surface and ground water (page 27)? I suggest a separate section for inclusion of specific ground water protection actions that need to be taken now (based on existing knowledge) and of priorities for continued monitoring. It should reference findings of both county and USGS sampling.

The USGS report mentions that some of the largest nitrogen concentrations were found near "animal pens" (page 21). This disclosure needs to be addressed in the county portion of the assessment. Where are these pens located? What size are they? What additional study and corrective action is needed?

Table 2 should note which set of data was taken by USGS and which was taken by Salt Lake County.

The reasoning for the absence of data for some parameters in Table 2 should be clarified. This apparent lack of data would contradict the statement at the bottom of page 3, and elsewhere in the assessment, which gives the impression that two sampling rounds were completed in which all of the same constituents were analyzed.

Under item 4 on page 42, it isn't clear what the term "accurate" means in the first sentence. The assessment needs to include and address the metals data from pages 21 and 22 of the USGS report.

The discussion of organic compounds (item 5 on page 42) states that "elevated levels... do not appear to occur at elevated levels...". This needs to be clarified and be more definitive. It needs to be consistent with USGS findings on page 24 of their report. The county assessment needs to be strengthened by discussing the relationship of detected contaminates to public health.

We commend the inclusion of the site-specific ground water studies underway (pages 44-61).

On page 51, the following partial sentence has a typographical error for correction: "(mill tailings with varying23egrees of activity)".

A map of the entire area with the plumes of contamination and their proximity to sensitive areas, such as drinking water sources, and wildlife or recreational areas is recommended, if not now, later after more study and funding.



United States Department of the Interior

GEOLOGICAL SURVEY

WATER RESOURCES DIVISION
Room 1016 Administration Building
1745 West 1700 South
Sait Lake City, Utah 84104

April 9, 1984

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SALE ALL THE TALE SHALLTY

Steven F. Jensen
Water Quality Program Manager
Salt Lake County Division of
Flood Control and Water Quality
2033 South State Street
Salt Lake City, UT 84115

Dear Mr. Jensen:

I have reviewed the draft copy of the "Assessment of the Shallow Aquifer in Salt Lake Valley" as you requested in your letter of March 30.

The report needs editorial review to correct spelling, grammatical, and stylistic errors. The style is inconsistent and varies from very formal to extremely colloquial (e.g. On p. 21 it is stated that "Virtually any old landfill is suspect.").

In the section discussing the results of the chemical analyses I am left with the feeling you are glossing over some of the more significant results.

- 1) pH. On page 31 it is stated that pH ranged from 6.0 to 8.0 and it seems to me you imply that this is well within the range of the expected. A cursory examination of hundreds of chemical analyses of ground water in the principal aquifer in Salt Lake County reveals only four pH values less than 7.0 and only one less than 6.9. Table 2 of the report shows that 18 of the 32 wells in the shallow aquifer had measured pH's less than 7.0. This may indicate widespread contamination of the shallow aquifer by either organic or mineral acids.
- 2) Contamination by organic chemicals. On page 42 you state "Elevated levels of organic parameters...do not appear to occur at elevated levels in the shallow aquifer observation wells." What is the basis for this statement? Is it based on the detection limits of 999 ug/L for organic chemicals used by the Utah State Lab? With such high detection limits even gross contamination of the aquifer could go undetected. I feel it is a serious error to imply that contamination of the aquifer by organic chemicals is not a problem, especially since several organic compounds identified by the EPA as potentially carcinogenic were found at levels well above the detection limits of the USGS Central Lab. Table 2, for example, reports that 660 ug/L phenols, 400 ug/L benzene and several different types of polychlorinated hydrocarbons were found.

3) Quality Assurance Data. Why is there no discussion of the data in Table 3? As an example, among all the samples analyzed for phenols by the Utah State Lab, the highest levels detected were in the field blank. Furthermore, there is no explanation of what the headings in Table 3 mean. What is a VOA7?

I believe the direction of ground-water movement shown in Figure 18 is wrong. Water level data for wells (B-1-1)26bad-1 and (B-1-1)26cda-1 support the conclusion that flow is more nearly east to west than shown. Thus all your conclusions on page 45 about how water quality changes as it moves through the refinery may be based on a misinterpretation of the water-level data.

The USGS report placed as appendix A at the end of the Salt Lake County report is a draft version. Since the USGS report is now available in final form I feel it would be appropriate to use the final version rather than draft version as an appendix.

Thank you for the opportunity to review the report. This compendium of chemical and water-level data should be very useful to water managers, contractors, and anyone else needing information on the shallow aquifer.

Ralph Seiler Hydrologist SALT'LAKE: CHIY CORPORATION

LEROY W. HOOTON, JR. DIRECTOR

JOSEPH S. FENTON SUPERINTENDENT, WATER RECLAMATION

WENDELL E. EVENSEN, P.E.
SUPERINTENDENT
WATER SUPPLY & WATERWORKS

DEPARTMENT OF PUBLIC UTILITIES

WATER SUPPLY & WATERWORKS WATER RECLAMATION 1530 SOUTH WEST TEMPLE SALT LAKE CITY, UTAH 84115

May 7, 1984

TED L. WILSON

RECEIVED

1-87 0.9 1984

Mr. Steven F. Jensen Water Quality Program Manager 2033 South State Street Salt Lake City, UT 84115

SALT LAKE COURTY FLOOD CONTIGUE & WATER QUALITY

Dear Steve:

RE: Review and Comment Request for Shallow Aquifer Assessment

Thank you for the opportunity of reviewing your report "Assessment of the Shallow Aquifer in Salt Lake Valley." Sorry that my comments are reaching you a little beyond the May 1, 1984, closing date.

In general, this is a very valuable report. I was especially glad to see the report zero in on specific sites and evaluate them for groundwater contamination in the shallow aquifer. My recommendation for the final report would be to have some comparison between the levels of contamination found and the maximum contaminant levels established in the National Interim Primary Drinking Water Regulation. This would be extremely meaningful. I noticed that some reference was made to those established contaminant levels in the incorporated, unpublished report by Seiler and Waddell. Perhaps they could be also be shown alongside the data obtained from your shallow wells.

Sincerely,

EROY W. HOOTON JR.

Director

WEE:ww 38:33:2



1596 West North Temple • Salt Lake City, UT 84116 • 801-533-9333

April 5, 1984

BESPENDED

F - 1 1 1984

FLOOD CONTROL & WATER AUALITY

Mr. Steven F. Jensen
Water Quality Program Manager
Salt Lake County Public Works Department
Flood Control and Water Quality Division
2033 South State Street
Salt Lake City, UT 84115

Dear Mr. Jensen:

We have reviewed the document "Assessment of The Shallow Aquifer in Salt Lake Valley", and we note that fish and aquatic habitat are listed as items dependent on water quality in the valley. We also note that waterfowl and wetland qualities are mentioned as being items which reflect water quality conditions. Even though premium fisheries do not occur within the Salt Lake Valley, the potential exists if water quality in the lower reaches of valley streams can be improved. The importance of our Waterfowl Management Areas and other wetlands around the Great Salt Lake needs no documentation and water quality will directly affect the quality of those areas, dependent wildlife and the resultant recreational opportunities to a great number of Utahns.

In addition, we concur that one of the basic criteria for the Shallow Aquifer Monitoring Networks (page 16) is to target potential non-point source contaminant plumes.

We also believe any reduction in underground and aboveground pollution will enhance the survival and presence of many species of urban wildlife found within the riparian areas of all Salt Lake Valley streams. We commend you for this assessment as a good start on what could become a critical future issue: Improved water quality in Salt Lake Valley.

We appreciate the opportunity to review and comment on this document.

Sincerely,

Douglas F. Day, Director

DIVISION OF WILDLIFE RESOURCES